

BRADYKININ B₁ RECEPTOR ANTAGONISTSCross Reference to Related Applications

- This application is a continuation of PCT application PCT/US00/19185 filed July 14, 2000, and published under PCT Article 21(2) in English as WO 01/05783
- 5 on January 25, 2001. PCT/US00/19185 claimed the priority of US provisional application 60/143,990, filed July 15, 1999. The entire disclosures of both are incorporated herein by reference.

Field of the Invention

- The invention relates to pyrimidines, triazines, and anilines that are
- 10 bradykinin B₁-receptor antagonists. The compounds are useful for treating diseases associated with inappropriate or excessive bradykinin receptor activity, such as diabetic vasculopathy, inflammation, pain, hyperalgesia, asthma, rhinitis, septic shock, atherosclerosis and multiple sclerosis.

Background of the Invention

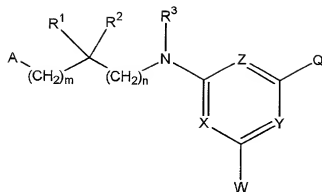
- 15 Bradykinin receptors of two classes are known. The B₁ receptor (B₁-BK) is not present in normal cells under normal conditions. In contrast, the B₂-BK receptor is normally present on many cell types or tissues. Although the B₁ receptor (B₁-BK) is not present under normal conditions, its synthesis is induced in blood vessel muscular layers during inflammation.
- 20 Recent reports point to an important role of bradykinin B₁ receptors in physiopathology. Dray and Perkins [*Trends in Neurosci.* **16**, 99-104(1993)] have reviewed the possible implication of B₁ receptors in various inflammatory states, in

5 tissue reactions and in hyperalgesia. Alvarez et al. [Clin. Sci.82, 513-519 (1992)]
have provided evidence that B₁ receptors are present in spontaneously hypertensive
rats (SHR), and Regoli et al. [PCT application WO 98/07746] have provided
evidence that inappropriate B₁ receptor activity is associated with some forms of
10 diabetes. In particular, it is known that capillary permeability is augmented in the
streptozotocin diabetic rat model, and the vascular BK receptors of the portal veins
of these animals have been shown to exhibit enhanced contractility and capillary
permeability in response to the B₁-agonist desArg⁸BK. This effect was abolished by
the B₁-antagonist Lys[Leu]desArg⁹BK while the B₂-antagonist HOE140 had no
15 effect. A similar increased sensitivity to desArg⁹BK was observed in untreated SHR
animals, prior to the establishment of hypertension, which was reversed by the same
B₁-antagonist. These results indicate that the B₁-receptor is a target for a drug-
preventive approach to diabetic or hypertensive vasculopathy.

15 Peptide antagonists of bradykinin receptors are known, although most
reported antagonists have activity towards B₂-receptors. There are to date very few
small molecule B₁ antagonists. It would be useful to have effective antagonists of
the B₁-BK receptor.

Summary of the Invention

20 In one aspect, the invention relates to a genus of bradykinin B₁ receptor
antagonists sharing the general formula I:



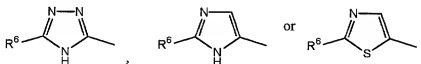
I

wherein:

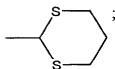
- (a) all of X, Y and Z are CH; or (b) one of X, Y and Z is N and the rest of X, Y and Z are CH; or (c) two of X, Y and Z are N and the other of X, Y and Z is CH; or (d) all of X, Y and Z are N;

A is A¹ or A²;

A¹ is R⁴R⁵N-C(O)-



- 10 A² is chosen from R⁷C(O)NH-, R⁷S(O)₂NH-, R⁴NH-, and R⁴O-;
Q is chosen from heteroaryl, aryl, -CH₂R¹³, -CH=N-OCH₃ and

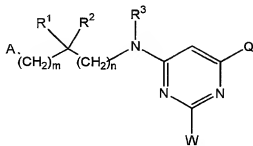


W is chosen from H, Cl, F, R⁸, C₁-C₄-alkylaryl, -OR⁸, -SR⁸, -NR⁹R¹⁰ and -NHC(O)R¹¹;

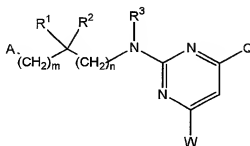
- 15 R¹ is chosen from alkyl, cycloalkyl, alkenyl, C₁-C₃-alkylcycloalkyl, heterocyclyl, C₁-C₃-alkylheterocyclyl, aryl, C₁-C₃-alkylaryl, heteroaryl, C₁-C₃-alkylheteroaryl, (C₁-C₃-alkyloxy)alkyl, (C₁-C₃-alkyloxy)cycloalkyl, (C₁-C₃-alkylthio)alkyl, (C₁-C₃-alkylthio)cycloalkyl and (C₁-C₃-alkylsulfonyl)alkyl;

- R^2 is H or C_1-C_3 -alkyl, or R^1 and R^2 taken together form a 5- to 7-membered ring structure optionally containing O, S or NR^{12} ;
- R^3 is H or C_1-C_6 -alkyl, or, when n is zero, R^2 and R^3 taken together may form a 6-membered ring, which may be fused to a six-membered saturated or aromatic carbocycle;
- R^4 is chosen from H, aryl, heteroaryl, C_1-C_4 -alkylaryl and C_1-C_4 -alkylheteroaryl,
- R^5 is H or C_1-C_3 -alkyl, with the proviso that both R^3 and R^5 cannot be alkyl;
- R^6 is aryl;
- R^7 is aryl or C_1-C_3 -alkylaryl;
- R^8 is chosen from alkyl, aryl, heteroaryl, substituted alkyl, C_1-C_4 -alkylaryl, C_1-C_4 -alkylheterocyclyl and C_1-C_4 -alkylheteroaryl;
- R^9 is chosen from H, alkyl, alkenyl, substituted alkyl, cycloalkyl, fluoroalkyl, C_1-C_4 -alkylcycloalkyl, $(C_1-C_4$ -alkoxy)alkyl, $(C_1-C_4$ -alkoxycarbonyl)alkyl, $(C_1-C_4$ -alkylthio)alkyl, heterocyclyl, C_1-C_4 -alkylheterocyclyl, C_1-C_4 -alkylaryl, C_1-C_4 -alkylheteroaryl, aryl and heteroaryl;
- R^{10} is H or C_1-C_3 -alkyl, or
- R^9 and R^{10} taken together may form a 5- to 7-membered ring structure optionally containing O, S, SO, SO_2 or NR^{12} , said ring optionally substituted with -OH, -CN, -COOH or -COOCH₃;
- R^{11} is aryl;
- R^{12} is chosen from H, C_1-C_3 -alkyl, alkoxycarbonyl, methoxyacetyl and
- R^{13} is chosen from -OH, -OTHP, 1-imidazolyl, and 1-pyrrolyl;
- m is zero or one; and
- n is zero or one, with the proviso that when A is A^2 , m and n cannot both be zero.

This genus may be considered to comprise subgenera of pyrimidines (IIa-IIc), triazines (III), anilines (IV) and pyridines (not shown):

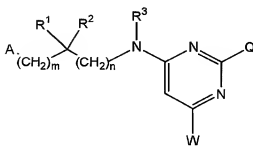


IIa

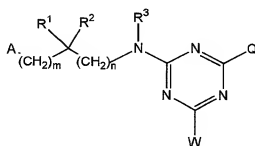


IIb

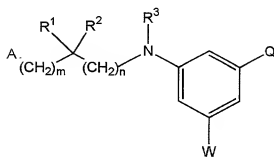
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IIc



III



IV

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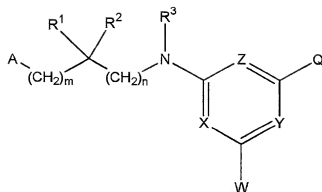
In another aspect, the invention relates to a method of treating a condition resulting from inappropriate bradykinin receptor activity comprising administering to a subject in need of such treatment a therapeutically effective amount of a compound of formula I. Conditions resulting from inappropriate bradykinin receptor activity include diabetic vasculopathy, post-capillary resistance or diabetic symptoms associated with insulinitis, inflammation, edema, liver disease, asthma, rhinitis, septic shock, pain, hyperalgesia, multiple sclerosis, atherosclerosis, Alzheimer's disease or closed head trauma. Of particular importance are chronic pain, pain associated with inflammation and dental pain. Diabetic symptoms associated with insulinitis include hyperglycemia, diuresis, proteinuria and increased nitrite and kallikrein urinary excretion. Stimulating hair growth or preventing hair loss may also be accomplished by administering to a subject in need of such treatment a therapeutically effective amount of a compound of formula I.

15 In another aspect, the invention relates to pharmaceutical compositions comprising a pharmaceutically acceptable carrier and compounds of formula I.

In another aspect, the invention relates to pharmaceutical compositions comprising a pharmaceutically acceptable carrier and compounds of formula I. The formulations may additionally comprise steroidal or nonsteroidal anti-inflammatory drugs (NSAIDS), cyclo-oxygenase (COX) inhibitors or selective cyclooxygenase-2 (COX-2) inhibitors.

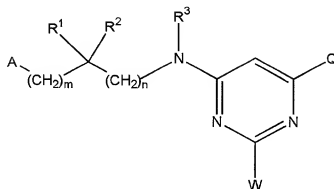
Detailed Description of the Invention

Preferred compounds of the invention are found in the class of pyrimidines of formula II



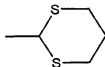
II

- These are compounds of formula I in which two of X, Y and Z are N and the third is CH. Three classes of pyrimidines can be limned, depending on which of X, Y and Z is CH. The first of these is the 4-pyrimidinamines, in which Z is CH. These have the formula IIa

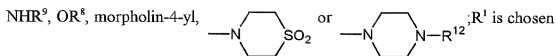


IIa

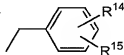
- In preferred embodiments, Q is chosen from imidazolyl, methylimidazolyl, pyrrolyl, methylpyrrolyl, pyrazolyl, methylpyrazolyl, furanyl, methylfuranyl, thienyl, oxazolyl, thiazolyl, pyridinyl, quinolinyl, 1-methylpyrimidin-2-onyl, phenyl, fluorophenyl, hydroxymethyl, tetrahydropyranyloxymethyl, imidazolylmethyl, pyrrolylmethyl, -CH=N-OCH₃, and

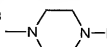


In particularly preferred embodiments Q is pyrrol-1-yl, imidazol-1-yl, furan-3-yl, 2-methylimidazol-1-yl or 4-methylimidazol-1-yl; A is R⁴R⁵N-C(O)-; W is Cl, R⁸,

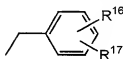


- from alkyl, cycloalkyl, C₁-C₃-alkylaryl, C₁-C₃-alkylcycloalkyl, C₁-C₃-alkylheterocyclyl, and C₁-C₃-alkylheteroaryl; R², R³ and R⁵ are H; R⁴ is C₁-C₄-alkylaryl or C₁-C₄-alkylheteroaryl; R⁸ is C₁-C₄-alkylaryl; R⁹ is chosen from
- 5 hydrogen, alkyl, fluoroalkyl, cyanoalkyl, hydroxy- and dihydroxyalkyl, (C₁-C₄-alkoxy)alkyl, (C₁-C₄-alkylthio)alkyl, C₁-C₄-alkylcycloalkyl, C₁-C₄-alkylaryl, heterocyclyl, C₁-C₄-alkylheteroaryl, and C₁-C₄-alkylheterocyclyl; and m and n are zero. When W is NHR⁹, preferred values of R⁹ are hydrogen; methyl; ethyl; 2,2,2-trifluoroethyl; allyl; cyclopropyl; 2-cyanoethyl; propargyl; methoxyethyl;
- 10 cyclopropylmethyl; (methylthio)ethyl; 3-methoxypropyl; 2-(3-pyridyl)ethyl; 2-(2-pyridyl)ethyl; 3-pyridylmethyl; 4-pyridylmethyl; sulfolan-3-yl; 3-tetrahydrofuran-2-yl; 2-tetrahydrofuran-2-ylmethyl; 3-(1-imidazolyl)propyl; 1-*t*-butoxycarbonyl-4-piperidylmethyl; and



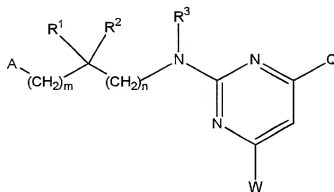
- SO₂NH₂, CF₃, COOCH₃, OCH₃, SO₂CH₃, N(CH₃)₂ or COOH; and R¹⁵ is chosen
- 15 from H, OCH₃, F and Cl. When W is , R¹² is preferably *t*-

- butoxycarbonyl, methoxyacetyl or phenyl. In another preferred embodiment of formula IIa, A is R⁴R⁵N-C(O)-; R¹ is chosen from *n*-butyl; cyclohexylmethyl; 2-methylpropyl; 3-methyl-1-butyl; cyclohexyl; 2,2-dimethylpropyl; benzyl; 2-thienylmethyl; 1-*t*-butoxycarbonyl-4-piperidyl; 4-chlorobenzyl; 2-pyranylmethyl;
- 20 4-pyranylmethyl; 4-pyranol and 1,1-dimethylethyl; R², R³ and R⁵ are H; R⁴ is aryl, pyridinylmethyl, pyridinyl or



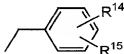
SO₂NH₂, CF₃, CH₃, COOCH₃, OCH₃, SO₂CH₃, N(CH₃)₂ or COOH; and R¹⁷ is H, OCH₃ or Cl. In these compounds, the carbon to which R¹ and R² are attached is preferably of the R absolute configuration, i.e. derivatives of D-amino acids, when m and n are zero.

- 5 In a second class of pyrimidines, the 2-pyrimidinamines, Y is CH. These have the formula IIb:



IIb

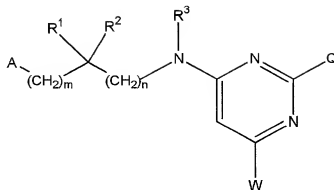
- Preferred embodiments are as for IIa. Particularly preferred embodiments are those in which Q is imidazolyl, pyrrolyl, pyridinyl, fluorophenyl or 2-thienyl. In these compounds, A is preferably R⁴R⁵N-C(O)-; W is H, Cl, NHR⁹ or OR⁸; R¹ is alkyl or C₁-C₃-alkylcycloalkyl; R², R³ and R⁵ are H; R⁴ is C₁-C₄-alkylaryl or C₁-C₄-alkylheteroaryl; R⁸ is C₁-C₄-alkylaryl; R⁹ is hydrogen, alkyl, fluoroalkyl, (C₁-C₄-alkoxy)alkyl, (C₁-C₄-alkylthio)alkyl, C₁-C₄-alkylcycloalkyl, C₁-C₄-alkylaryl, heterocyclyl, C₁-C₄-alkylheteroaryl, or C₁-C₄-alkylheterocyclyl; and m and n are zero. Among these, the most preferred compounds are those in which W is NHR⁹ and R⁹ is



COOCH₃, OCH₃, SO₂CH₃, N(CH₃)₂ or COOH; and R¹⁵ is H, OCH₃ or Cl.

In the third class of pyrimidines, a different set of 4-pyrimidinamines, X is

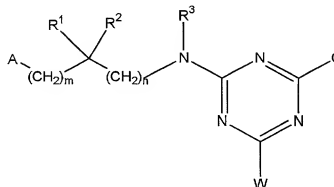
CH. These have the formula IIc:



IIc

Preferred embodiments are as for IIa. Particularly preferred embodiments are those in which Q is imidazolyl or pyrrolyl. In these compounds, A is preferably R⁴R⁵N-C(O)-; W is NHR⁶; R¹ is cyclohexylmethyl; 2-methylpropyl or 3-methyl-1-butyl; R², R³ and R⁵ are H; and R⁴ and R⁶ are benzyl or substituted benzyl.

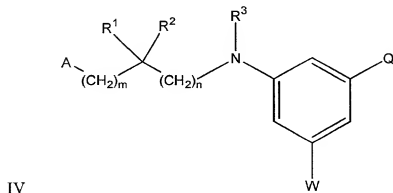
Triazines form another subgenus of the invention according to formula I; in this subgenus, all of X, Y, and Z are N. The triazines of interest have the formula III



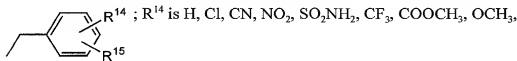
III

Preferred embodiments are as for the pyrimidines. Particularly preferred embodiments are those in which Q is imidazolyl or pyrrolyl. In these compounds, A is preferably R⁴R⁵N-C(O)-; W is NHR⁶; R¹ is cyclohexylmethyl; 2-methylpropyl or 3-methyl-1-butyl; R², R³ and R⁵ are H; and R⁴ and R⁶ are benzyl or substituted benzyl.

Anilines form another subgenus of the invention according to formula I in which all of X, Y, and Z are CH. Anilines of the invention have the formula IV:



Preferred embodiments are as for the pyrimidines. Particularly preferred
 5 embodiments are those in which Q is imidazolyl or pyrrolyl. In these compounds, A is preferably R⁴R⁵N-C(O)-; W is NHR⁹; R¹ is alkyl, cycloalkyl, C₁-C₃-alkylaryl or C₁-C₃-alkylcycloalkyl; R², R³ and R⁵ are H; R⁴ is C₁-C₄-alkylaryl; R⁹ is



- 10 “Alkyl” is intended to include linear, or branched hydrocarbon structures and combinations thereof; hydrocarbons of 20 or fewer carbons are generally preferred. “Lower alkyl” means alkyl groups of from 1 to 6 carbon atoms. Examples of lower alkyl groups include methyl, ethyl, propyl, isopropyl, butyl, s- and t-butyl, pentyl, hexyl, and the like.

- 15 “Cycloalkyl” includes cycloalkyl groups of from 3 to 12 carbon atoms. Examples of “cycloalkyl” groups include c-propyl, c-butyl, c-pentyl, c-hexyl, 2-methylcyclopropyl, cyclopropylmethyl, cyclopentylmethyl, norbornyl, adamantyl, myrtanyl and the like.

“Alkenyl” refers to a C_2 to C_{20} hydrocarbon of a linear, branched, or cyclic (C_5 - C_6) configuration, and combinations thereof, having one or two degrees of unsaturation. C_2 - C_8 Alkenes are preferred. Examples of alkenyl groups include vinyl, allyl, isopropenyl, pentenyl, hexenyl, c-hexenyl, 1-propenyl, 2-butenyl, 2-methyl-2-butenyl, 2,4-hexadienyl and the like.

Alkynyl is C_2 - C_8 alkynyl of a linear or branched configuration and combinations thereof. Examples of alkynyl groups include ethyne, propyne, butyne, pentyne, 3-methyl-1-butyne, 3,3-dimethyl-1-butyne, and the like.

C_1 to C_{20} Hydrocarbon includes alkyl, cycloalkyl, alkenyl, alkynyl, aryl and combinations thereof. Examples include phenethyl, cyclohexylmethyl and naphthylethyl.

“Alkoxy” means alkoxy groups of from 1 to 6 carbon atoms of a straight, branched, cyclic configuration and combinations thereof. Examples of alkoxy groups include methoxy, ethoxy, propoxy, isopropoxy, cyclopropyloxy, cyclohexyloxy, and the like.

Halogen includes F, Cl, Br, and I, with F and Cl as the preferred groups. “Halophenyl” means phenyl substituted by 1-5 halogen atoms. Halophenyl includes pentachlorophenyl, pentafluorophenyl, and 2,4,6-trichlorophenyl. “Fluoroalkyl” refers to an alkyl residue in which one or more hydrogen atoms are replaced with F, for example: trifluoromethyl, difluoromethyl, and pentafluoroethyl, 2,2,2-trifluoroethyl.

“Aryl” and “heteroaryl” mean a 5- or 6-membered aromatic or heteroaromatic ring containing 0-3 heteroatoms selected from O, N, and S; a bicyclic 9- or 10-membered aromatic or heteroaromatic ring system containing 0-3 heteroatoms selected from O, N, and S; or tricyclic 13- or 14-membered aromatic

or heteroaromatic ring system containing 0-3 heteroatoms selected from O, N, and S; each of which rings is optionally substituted with up to three substituents chosen independently from lower alkyl, =O, nitro, halogen, hydroxy, alkoxy, alkylsulfonyl; methylenedioxy, alkoxyethoxy, cyano, amino, alkylamino, dialkylamino, acylamino, aminosulfonyl, C₁-C₆-alkoxycarbonyl, carboxy, methylsulfonamido, perfluoroalkyl, phenyl, benzyl, trityl, and phenoxy. 6- to 14-Membered aryl residues include, for example, benzene and naphthalene, and the 5- to 10-membered heteroaryl residues include, for example, imidazole, pyridine, indole, oxazole, thiophene, benzopyranone, benzodioxan, benzodioxole, thiazole, furan, benzimidazole, quinoline, isoquinoline, quinoxaline, pyrimidine, pyrimidinone, pyridazine, tetrazole, and pyrazole.

“Arylalkyl” and “alkylaryl” denote an aryl residue attached to the parent structure through an alkyl residue. The alkyl need not be straight chain. Examples include benzyl, phenethyl, 2-phenylpropyl, 4-chlorobenzyl, and the like. The alkyl may also be a fused cycloalkyl such as indan (e.g. indan-2-yl), tetralin, and fluorene (e.g. fluoren-9-yl) or a substituted alkyl, such as in 1-hydroxyindan-2-yl. “Heteroarylalkyl” denotes a residue comprising an alkyl attached to a heteroaryl ring such as pyridinylmethyl, pyrimidinylethyl, and the like.

“Heterocycloalkyl” means a cycloalkyl where one to three carbon atoms is replaced with a heteroatom, such as O, NR (R= H, alkyl), N→O, S, SO, SO₂ and the like. The term includes residues in which one or more rings is optionally substituted with up to three substituents chosen independently from lower alkyl, =O, halogen, hydroxy, alkoxy, amino, alkylamino, dialkylamino, acylamino, aminosulfonyl, C₁-C₆-alkoxycarbonyl, carboxy, methylsulfonamido, perfluoroalkyl, phenyl, benzyl, trityl, and phenoxy. When two heteroatoms are separated by a single carbon, the resulting heterocycloalkyls tend to be unstable in aqueous solutions and are therefore not preferred. Examples of heterocycloalkyls

include: tetrahydrofuran, tetrahydropyran, piperidine, pyridine-N-oxide, 2-methyl-1,3-dithiane, dioxane, and the like.

“Substituted” alkyl, alkenyl, cycloalkyl, aryl, heteroaryl or heterocycloalkyl means alkyl, alkenyl, cycloalkyl, aryl, heteroaryl or heterocycloalkyl, wherein

- 5 hydrogen atoms are replaced by halogen, hydroxy, carboxy, carboalkoxy, carboxamido, cyano, carbonyl, hydroxyimino, alkoxyimino, nitro, alkoxy, methylenedioxy, alkoxyethoxy, amino, alkylamino, dialkylamino, acylamino, aminosulfonyl, C₁-C₆-alkoxycarbonyl, methylsulfonamido, methylsulfonyl, methylthio, perfluoroalkyl, phenyl, benzyl, trityl, phenoxy, amidino, guanidino, ureido, and benzyloxy.
- 10

Abbreviations and Definitions

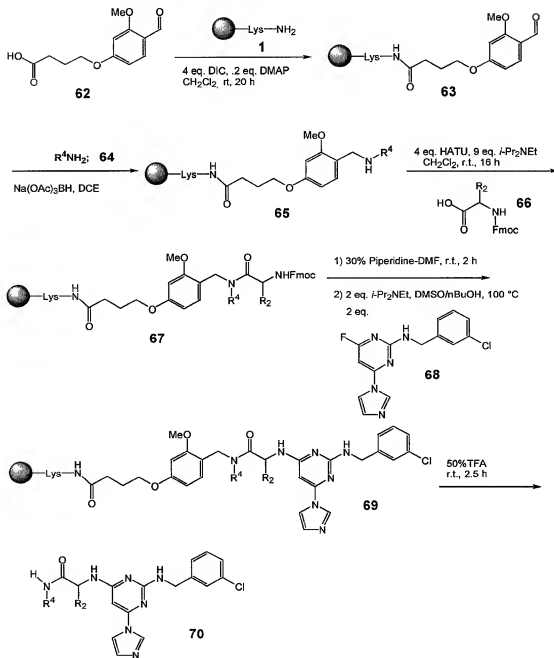
The following abbreviations and terms have the indicated meanings throughout:

- | | | | |
|----|------|---|--|
| | Ac | = | acetyl |
| 15 | BNB | = | 4-bromomethyl-3-nitrobenzoic acid |
| | Boc | = | t-butyloxy carbonyl |
| | Bu | = | butyl |
| | c- | = | cyclo |
| | DBU | = | diazabicyclo[5.4.0]undec-7-ene |
| 20 | DCM | = | dichloromethane = methylene chloride = CH ₂ Cl ₂ |
| | DEAD | = | diethyl azodicarboxylate |
| | DIC | = | diisopropylcarbodiimide |
| | DIEA | = | N,N-diisopropylethyl amine |
| | DMAP | = | 4-N,N-dimethylaminopyridine |
| 25 | DMF | = | N,N-dimethylformamide |
| | DMSO | = | dimethyl sulfoxide |
| | DVB | = | 1,4-divinylbenzene |

	EEDQ =	2-ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline
	Fmoc =	9-fluorenylmethoxycarbonyl
	GC =	gas chromatography
	HATU =	O-(7-Azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium
5		hexafluorophosphate
	HOAc =	acetic acid
	HOBt =	hydroxybenzotriazole
	Me =	methyl
	mesyl =	methanesulfonyl
10	MTBE =	methyl t-butyl ether
	NMO =	N-methylmorpholine oxide
	PEG =	polyethylene glycol
	Ph =	phenyl
	PhOH =	phenol
15	PfP =	pentafluorophenol
	PPTS =	pyridinium p-toluenesulfonate
	PyBroP =	bromo-tris-pyrolidino-phosphonium hexafluorophosphate
	rt or RT =	room temperature
	sat'd or sat. =	saturated
20	s- =	secondary
	t- =	tertiary
	TBDMS=	t-butyldimethylsilyl
	TFA =	trifluoroacetic acid
	THF =	tetrahydrofuran
25	TMOF =	trimethyl orthoformate
	TMS =	trimethylsilyl
	tosyl =	p-toluenesulfonyl
	Trt =	triphenylmethyl

The compounds of the invention are synthesized as follows.

Scheme 1 Generic Solid Phase Synthesis

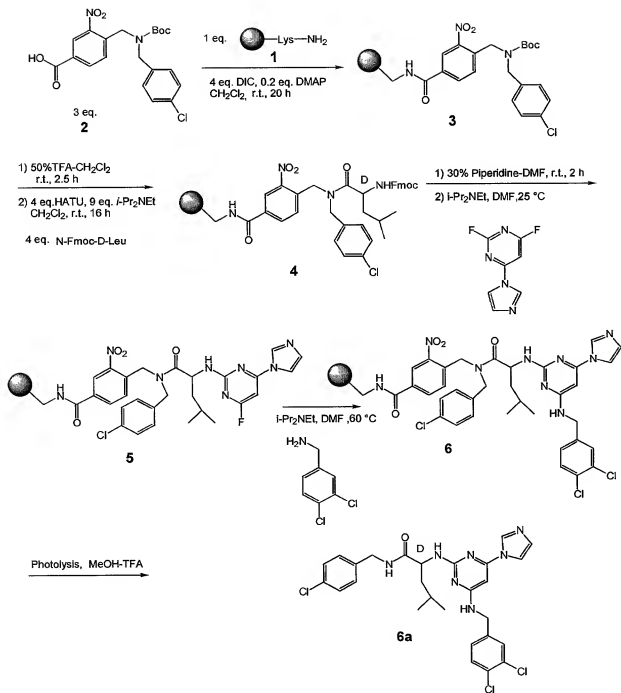


Amino functionalized TentaGel resin **1** (10 g 5.2 mmole) was suspended in 50 mL of CH_2Cl_2 and treated with 3.73 g of linker acid **62** (15.6 mmole), 3.25 mL of DIC (20.8 mmole), and 63 mg of DMAP (0.52 mmole). After 48h at room temperature, 3.77g of linker acid **62**, 3.25 mL of DIC and 2.1g HOBt were added.

- 5 The mixture was shaken at room temperature for 17 h and then washed with DMF twice, CH_2Cl_2 ten times to give resin **63**. The resins **63** was treated with amine R^4NH_2 **64** and $\text{Na}(\text{OAc})_3\text{BH}$ in dichloroethane at room temperature for 36h then washed with methanol 5 times and methylene chloride 5 times to give resin-bound amine **65**. The amine was coupled with an N-Fmoc amino acid (**66**) by treatment
- 10 with HATU and $i\text{-Pr}_2\text{NEt}$ in methylene chloride at room temperature for 48 h to provide resin **67**. Fmoc on resin **67** was removed by treatment with 30% piperidine in DMF and the resulting resin-bound amine was then reacted with fluoropyrimidine **68**, $i\text{-Pr}_2\text{NEt}$ in $\text{DMSO}:\text{nBuOH}$ (1:1) at 100 °C for 18 h and then
- 15 washed with methanol, CH_2Cl_2 to give resin bound product **69**. The final product was cleaved off resin by treatment with TFA for 3 h to give product **70**.

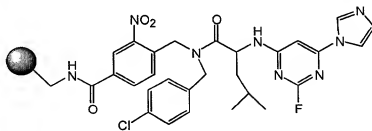
- The fluoropyrimidine **68** was prepared by stirring together 315 mg 6-imidazolyl-2,4-difluoropyrimidine (1.7 mmole), 265 mg of 3-chlorobenzylamine and 0.5 mL of $i\text{-Pr}_2\text{NEt}$ in 30 mL of THF at 50 °C for 16 h, then cooling to room temperature. The reaction was diluted with ethyl acetate and washed with saturated
- 20 NH_4Cl , H_2O , brine, dried over MgSO_4 and concentrated. The crude product was purified by flash chromatography (eluted with 4:5:1 EtOAc : hexanes : MeOH) to give 160mg of **68** (more polar product as compared the other regioisomer).

Scheme 2



Scheme 2 depicts a similar synthesis to that of Scheme 1, except the linker is photolytically cleavable instead of acid cleavable. As shown in Scheme 2, 2.5 g of amino functionalized TENTAGEL™ resin **1** (0.70 mmole) was suspended in 10 mL of CH_2Cl_2 and treated with 0.882 g of linker acid **2** (2.1 mmole), 0.44 mL of DIC (2.8 mmole), and 17 mg of DMAP (0.14 mmole). The mixture was shaken at room temperature for 17 h and then washed with CH_2Cl_2 ten times to give resin **3**.

1.13 g of resin **3** was treated with 50% TFA- CH_2Cl_2 at room temperature for 1.5 h and then washed with CH_2Cl_2 ten times, 15% Et_3N - CH_2Cl_2 for 10 min, and CH_2Cl_2 for 5 times. The deprotected resin was then suspended in 12 mL of CH_2Cl_2 and treated with 449 mg of N-Fmoc-D-Leu (1.27 mmole), 483 mg of HATU (1.27 mmole), and 0.50 mL of $i\text{-Pr}_2\text{NEt}$ (2.85 mmole). The mixture was shaken for 19 h at ambient temperature and then washed 5 times to give resin **4**. Fmoc on resin **4** was removed by treatment with 30% piperidine in DMF and the resulting resin-bound amine (0.32 mmole) was then reacted with 182 mg of 6-imidazolyl-2,4-difluoropyrimidine (0.64 mmole), 0.34 mL of $i\text{-Pr}_2\text{NEt}$ (1.92 mmole) in 10 mL of DMF at 23°C for 17 h and then washed with DMF, CH_2Cl_2 to give resin **5**. This reaction also produces the other regioisomer **5a**,

**5a**

which provides entry into the series of pyrimidines of general formula IIa above. The two are separated after cleavage. For simplicity, only the further transformations in the IIb series are shown in Scheme 2. The resin-bound fluoride **5** was treated with 0.25 mL of 3,4-dichlorobenzylamine (1.6 mmole) in 15 mL of DMF and 0.30 mL of Hünig's base at 60°C for 18h and then cooled to room

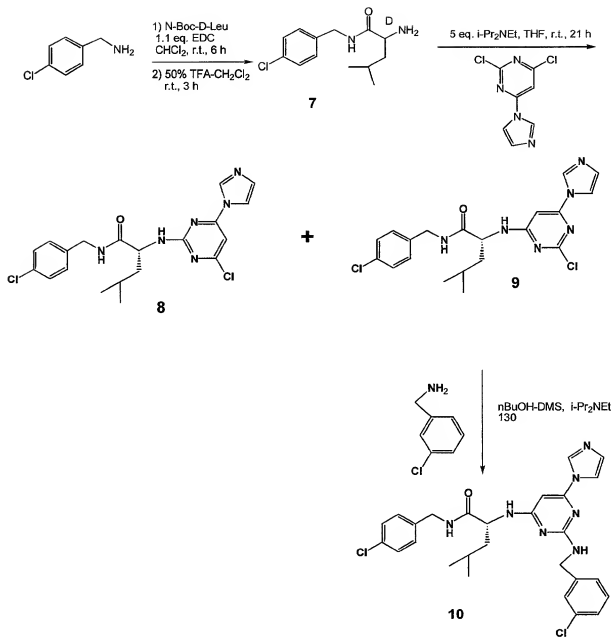
temperature and washed with DMF, CH_2Cl_2 . The final product was cleaved off resin by photolysis in MeOH for 17 h to give 49.2 mg of crude product.

Purification by flash chromatography (eluted with 5:5:1 EtOAc : hexanes : MeOH) gave 27.2 mg of **6a** (later determined to be mixture of two regioisomers with 1:1 ratio).

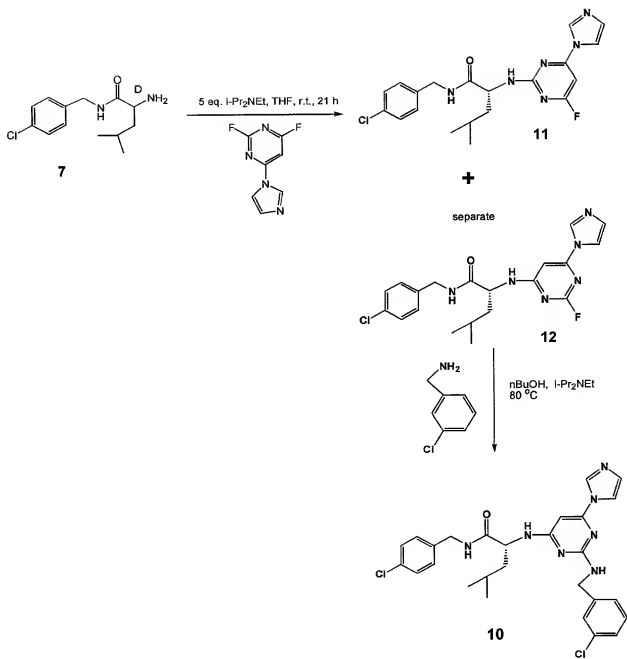
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Scheme 3



Scheme 4

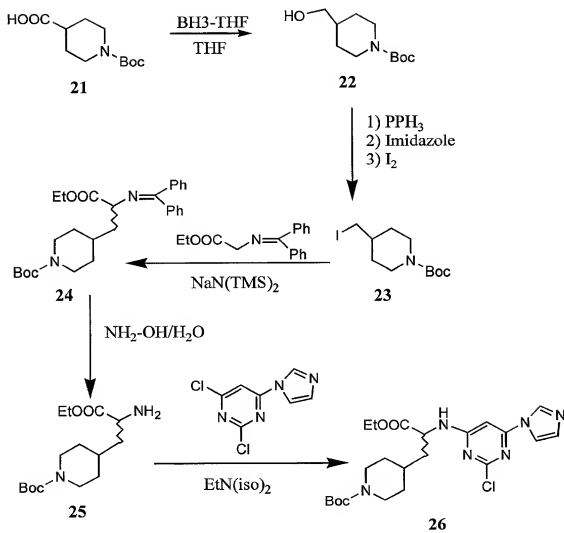


Scheme 3 illustrates a solution phase synthesis via chloropyrimidines and Scheme 4 illustrates a solution phase synthesis via fluoropyrimidines. As shown in Scheme 3, EDC (5.18 g, 26.47 mmole) was added into a solution of N-Boc-D-leucine (6.0 g, 24.07 mmole) in 250 mL of CH_2Cl_2 , followed by 2.99 mL of 4-chlorobenzylamine (24.07 mmol). The mixture was stirred at room temperature for 4 h then diluted with ethyl acetate and washed with 1 N HCl twice, saturated NaHCO_3 and brine twice, dried over MgSO_4 and concentrated to give 7.92 g of crude amide product which was treated with 50% TFA in CH_2Cl_2 at room temperature for 4 h. The solvent was removed and the residue was taken up into ethyl acetate and washed with 2 N NaOH aqueous solution, then brine, dried over MgSO_4 and concentrated to give amine product **7** quantitatively. Three hundred ninety milligrams of the free amine **7** (1.1 mmole) was treated with 0.6 mL of *i*-Pr₂NEt and 500 mg 6-imidazolyl-2,4-dichloropyrimidine (2.0 mmole) in DMF at 50 °C for 16 hr, then diluted with ethyl acetate and washed with saturated NH_4Cl , H_2O , brine, dried over MgSO_4 and concentrated and purification by flash chromatography (eluted with 8:10:1 EtOAc : Hexanes : MeOH) to give 200 mg of **8** and 130 mg of **9**. Ninety two milligrams of **9** (0.21 mmole) in 3 mL of *n*-butanol was treated with 0.9 mL of 3-chlorobenzylamine and 1 mL of *i*-Pr₂NEt at 100 °C for 16 h, then cooled to room temperature, diluted with ethyl acetate and washed with saturated NH_4Cl , H_2O , brine, dried over MgSO_4 and concentrated. The crude product was purified by flash chromatography (eluted with 4:5:1 EtOAc : Hexanes : MeOH) to give 97.2 mg of **10**.

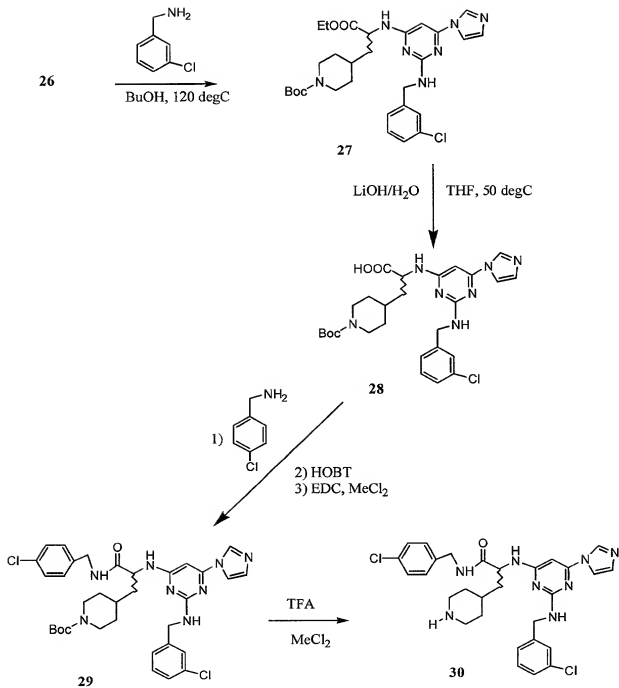
Alternatively, as illustrated in Scheme 4, 280 mg of the free amine **7** (1.1 mmole) was treated with 0.25 mL of *i*-Pr₂NEt and 200 mg of 6-imidazolyl-2,4-difluoropyrimidine (1.1 mmole) in THF at room temperature for 13 hr, then diluted with ethyl acetate and washed with saturated NH_4Cl , H_2O , brine, dried over MgSO_4 and concentrated. The crude product was purified by flash chromatography (eluted with 8:10:1 EtOAc : hexanes : MeOH) to give 35 mg of **11** (less polar product)

and 80 mg of **12** (more polar product). Four hundred fifty milligrams of **12** (1.08 mmole) in 50 mL of THF or n-butanol was treated with 1.7 g of 3-chlorobenzylamine and 5 mL of i-Pr₂NEt at 80 °C for 16 h then diluted with ethyl acetate and washed with saturated NH₄Cl, H₂O, brine, dried over MgSO₄ and
5 concentrated. The crude product was purified by flash chromatography (eluted with 6:12:1 EtOAc : hexanes : MeOH) to give 350 mg of **10**.

Scheme 5



Scheme 5 (continued)



Scheme 5 illustrates a synthesis of a member of the subgenus in which R¹ is heterocycloalkyl. According to Scheme 5 a dry 500 mL round bottom flask (oven-heated/argon cooled), was charged with 25 g (109.2 mmol) of Boc-isonipecotic acid (**21**). The flask was purged with argon, and 150 mL of dry THF were injected by syringe into the air-free system. The mixture was then stirred while being cooled to 0°C, and an oil-bubbler was attached, then 131 mL of a 1 M solution of borane/THF (131 mmol) were injected into the solution slowly, and the solution was stirred for ½ hour. Methanol was dripped into the solution slowly until bubbles ceased to be evolved. The solution was washed with 200 mL of a saturated sodium bicarbonate solution, and extracted twice with ethyl acetate, and the organic layer was dried over magnesium sulfate. The yield of the reaction was 22.44g (96%) of the **22** product as a white solid. ¹H NMR in CDCl₃: a 3H multiplet from 0.85-1.2 ppm, a 9H singlet at 1.45 ppm, a 4H multiplet 1.455-1.8 ppm, a 2H broad signal at 2.65 ppm, a 1H broad signal at 3.45 ppm, and a 1H broad signal at 3.6 ppm.

A 250 mL round bottom flask was charged with 5.8 g (27 mmol) of **22**, 8.5 g (32.37 mmol) of triphenylphosphine, and 2.2 g (32.37 mmol) of imidazole. One hundred millileters of methylene chloride were added, and the resulting solution was stirred at 0°C for about 5 minutes. Finally, 8.2 g (32.37 mmol) of iodine were added and the solution was stirred at 0°C for 5 minutes and at room temp for about 1 hour. The reaction mixture was diluted with 200 mL of hexane, and the triphenylphosphine oxide precipitate was filtered off (this was repeated until all precipitate was removed). The crude mixture was purified by flash chromatography using a 5%-10% ethyl acetate/hexane solvent system. A Phosphomolybdic acid stain (PMA), was used to see the product on the TLC plate. The resulting yield of pure **23** as an oil was 2.6g (30%). ¹H NMR in CDCl₃: 2H quartet at 1.1 ppm (J=12 Hz), a 9H singlet at 1.4 ppm, a 1H broad signal at 1.55 ppm, a 2H doublet at 1.75

($J=12$ Hz), a 2H broad signal at 2.65 ppm, a 2H doublet at 3.05 ppm ($J=6$ Hz), and a 2H broad signal at 4.1 ppm. The $R_f=0.13$ using a 5% ethyl acetate/hexane solvent system.

5 A dry 250 mL round bottom flask (oven heated/argon cooled), was charged with 1.3 g (5.13mmol) of N-(diphenylmethylene) glycine ethyl ester. The flask was purged with argon, and 100 mL of dry THF were injected into the air-free system. The resulting solution was cooled to -78°C with stirring, and 6.2 mL (6.15mmol) of a 0.1 M solution of sodium hexamethyldisilazane in THF were injected into the solution. The reaction was stirred at -78°C for $\frac{1}{2}$ hr, and a solution of 2 g of **23** in 10 dry THF was injected into the system. The solution was stirred at -78°C for 1 hr, at 0°C for 1 hr, and at room temp overnight. The reaction mixture was washed with a solution of 1 g (6.15 mmol) of citric acid in water, and diluted with 200 mL of ethyl acetate. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography using a 10% ethyl 15 acetate/hexane solvent system. The yield was 1.45 g (61%) of solid product **24**. ^1H NMR in CDCl_3 : A 3H broad multiplet from 0.8-1.15 ppm, a 4H broad signal at 1.25 ppm, a 9H singlet at 1.4 ppm, a 2H broad signal at 1.5 ppm, a 1H broad triplet at 1.85 ppm, a 2H broad quartet at 2.6 ppm, a 2H broad signal at 3.95 ppm, a 2H broad signal at 4.15 ppm, a 2H triplet at 7.15 ppm ($J=3.6$ Hz), a 6H multiplet from 20 7.25-7.5 ppm, and a 2H doublet at 7.6 ppm ($J=9$ Hz). The $R_f=0.22$ using a 10% ethyl acetate/hexane solvent system. ESI MS at 465 MH^+ .

A 100 mL round bottom flask was charged with 0.35 g (0.75 mmol) of **24**, and 20 mL of ethanol were added to the flask. With stirring, 0.5 mL of a 50% (by weight) solution of hydroxylamine was added followed by 0.5 mL of glacial acetic acid (5 minutes later). The reaction was stirred for 10 minutes, until the starting 25 material disappeared by TLC. The reaction mixture was diluted with 100 mL ethyl acetate, 20 mL of a brine solution was added, followed by basification using 0.5 M

NaOH. The organic layer was extracted, and the aqueous layer was then extracted with two 20 mL portions of methylene chloride. The combined organic layers were dried over magnesium sulfate. The crude mixture was purified by flash chromatography using a 55% ethyl acetate/hexane solvent system. A ninhydrin stain was used to see the product spot on the TLC plate. The yield of pure **25** as an oil was 0.25 g (96%). ¹H NMR in CDCl₃: A 1H multiplet from 0.9-1.05 ppm, a 3H broad triplet at 1.1 ppm, a 3H triplet at 1.25 ppm (J=6 Hz), an 11H broad signal at 1.4 ppm, a 3H multiplet from 1.5-1.8 ppm, a 2H broad triplet at 2.7 ppm, a 3H quartet at 3.45 ppm (J=3.6 Hz), and a 4H multiplet from 4-4.2 ppm. The R_f= 0.22 using a 55% ethyl acetate/hexane solvent system.

A 50 mL round bottom flask was charged with 0.310 g (1 mmol) of **25** and 5 mL of DMF. With stirring, 0.22 g (1 mmol) of the pyrimidine/imidazole subunit, and 0.35 mL (2 mmol) of diisopropylethylamine (Hünig's base) were added. The mixture was stirred at 90°C overnight. The reaction mixture was diluted with 200 mL of ethyl acetate, and washed with water. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography using an 80%-90% ethyl acetate/hexane solvent system. The yield of the reaction was 0.14 g of the regio-isomer with substitution of the pyrimidine at the 2-position and 0.12 g (25%) of the desired regio-isomer **36** (oil), (total yield is 54%). ¹H NMR in CDCl₃: a 1H multiplet from 0.9-1.05 ppm, a 3H broad triplet at 1.15 ppm, a 2H triplet at 1.3 ppm (J=6Hz), a 9H singlet at 1.45 ppm, a 2H broad signal at 1.7 ppm, a 2H broad signal at 1.85 ppm, a 2H broad triplet at 2.65 ppm, a 2H broad signal at 4.1 ppm, a 2H quartet at 4.2 ppm (J=2.4 Hz), a 1H broad signal at 4.9 ppm, a 1H doublet at 6.05 ppm (J=9 Hz), a 1H broad singlet at 6.3 ppm, a 1H singlet at 7.15 ppm, a 1H singlet at 7.5 ppm, and a 1H singlet at 8.3 ppm. The R_f of the desired regio-isomer was about 0.22 using an 80% ethyl acetate/hexane solvent system. The pure product gave a molecular ion of 480, MH⁺.

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A 50 mL round bottom flask was charged with 0.12 g (0.25 mmol) of **26**, 0.142 g (1mmol) of 3-chlorobenzylamine, and 5 mL of dry n-butanol. The solution was stirred at 120°C overnight. The reaction mixture was diluted with 200 mL of ethyl acetate, and washed with water. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography using a 90%-95% ethyl acetate/hexane solvent system. The yield was 0.125 g (87%) of **27** as an oil. ¹H NMR in CDCl₃: a 1H triplet at 0.9 ppm (J=6), a 2H broad signal at 1.1 ppm, a 3H triplet at 1.25 ppm (J=4.8 Hz), a 9H singlet at 1.45 ppm, a 5H broad signal at 1.65 ppm, a 2H broad signal at 2.6 ppm, a 4H broad signal at 4.1 ppm, a 2H doublet at 4.55 ppm (J=6 Hz), a 1H broad signal at 4.7 ppm, a 1H doublet at 5.4 ppm (J=9 Hz), a 1H singlet at 5.75 ppm, a 1H singlet at 7.1 ppm, a 3H singlet at 7.2 ppm, a 1H singlet at 7.35 ppm, a 1H singlet at 7.5 ppm, and a 1H singlet at 8.25 ppm. The R_f of the product was about 0.28 using an 80% ethyl acetate/hexane solvent system. The pure product gave a molecular ion of 584, MH⁺.

A 50 mL round bottom flask was charged with 0.125 g (0.214 mmol) of **27** and 10 mL of THF. With stirring, a solution of 0.09 g (2.14 mmol) of lithium hydroxide in 10 mL of water was added. The solution was heated at 55°C for 2 hr. The reaction mixture was diluted with 200 mL of ethyl acetate, and washed with a solution of 0.412 g (2.14 mmol) of citric acid in water to neutralize the excess base present. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography using an 95% ethyl acetate/methanol solvent system. The yield was 0.1 g (83%) of pure **28** as a white solid. ¹H NMR in CDCl₃: a 1H broad signal at 0.9 ppm, a 3H broad signal at 1.1 ppm, a 2H triplet at 1.25 ppm (J=6 Hz), a 9H singlet at 1.4 ppm, a 4H broad signal at 1.65 ppm, a 2H broad signal at 2.45 ppm, a 3H broad signal at 4 ppm, a 2H broad signal at 4.3-4.8 ppm, a 1H broad signal at 5.85 ppm, a 1H singlet at 7.05 ppm, a 3H singlet at 7.15 ppm, a 1H doublet at 7.25 ppm (J=3.6), a 1H singlet at

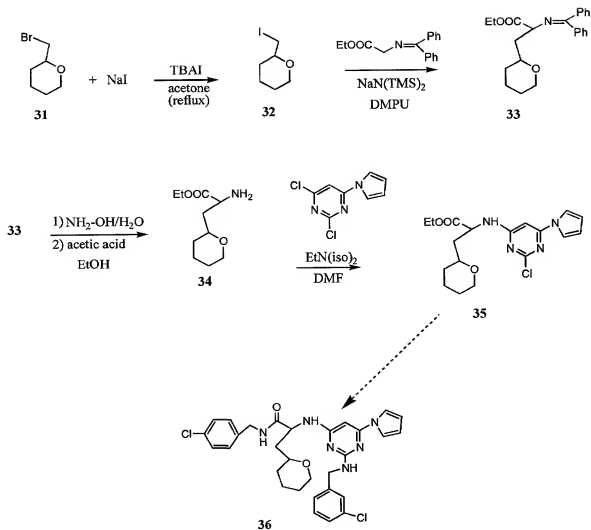
7.5 ppm, and a ¹H singlet at 8.5 ppm. The R_f of the product was about 0.08 using a 95% ethyl acetate/methanol solvent system. The pure product gave a molecular ion of 556, consistent with its molecular weight of 555, MH⁺.

5 A 50 mL round bottom flask was charged with 0.099 g (0.178 mmol) of **28** and 20 mL of methylene chloride. With stirring, 0.048 g (0.356 mmol) of 1-hydroxybenzotriazole (HOBt) and 0.068 g (0.356 mmol) of 1-(3-dimethylaminopropyl)-3-ethyl-carbodiimide hydrochloride (EDC), were added to the solution, then 1 mL of DMF was added to aid in solubility, and the solution was stirred for 20 minutes, until the acid intermediate spot disappeared by TLC. Fifty
10 milligrams (0.356 mmol) of 4-chlorobenzylamine was added to the solution and it was stirred for 2 hrs. The reaction mixture was diluted with 200 mL of ethyl acetate and washed successively with solutions of 0.5 M HCl, 0.5 M NaOH, and brine. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography using 100%-98% ethyl
15 acetate/methanol as the solvent system. The yield was 0.085 g (71%) of pure **29** as an oil. ¹H NMR in CDCl₃: a 4H multiplet from 0.9-1.3 ppm, a 9H singlet at 1.4 ppm, a 5H broad signal at 1.6 ppm, a 1H multiplet from 1.75-2.15 ppm, a 2H broad signal at 2.6 ppm, a 2H singlet at 3.85 ppm, a 1H broad signal at 4.05 ppm, a 4H multiplet from 4.3-4.6 ppm, a 1H doublet at 5.3 ppm (J=6), a 1H singlet at 5.7
20 ppm, a 2H singlet at 7.1 ppm, a 7H multiplet from 7.15-7.3 ppm, a 1H singlet at 7.45 ppm, and a 1H singlet at 8.25 ppm. The R_f of the product was about 0.24 using a 95% ethyl acetate/methanol solvent system. The pure product gave a molecular ion of 679, consistent with its molecular weight of 678 amu.

25 A 50 mL round bottom flask was charged with 0.020 g (0.03 mmol) of **29** and 3 mL of methylene chloride. With stirring, 1.5 mL (0.02 mmol) of trifluoroacetic acid was added, and the solution was stirred for about 20 minutes, until the Boc-containing intermediate disappeared by TLC. The reaction mixture

was diluted with 10 mL toluene and evaporated twice. The product **30** was diluted with 50 mL ethyl acetate, and washed with 0.5 M NaOH. ^1H NMR in CDCl_3 : a 5H multiplet from 0.75-1 ppm, a 5H multiplet from 1.5-1.8 ppm, a 1H singlet at 1.95 ppm, a 2H quartet at 2.6 ppm ($J=14$), a 1H broad signal at 3.1 ppm, a 2H singlet at 3.65 ppm, a 4H multiplet from 4.-4.7 ppm. A 1H singlet at 5.9 ppm, a 9H multiplet from 7.05-7.15 ppm, a 1H singlet at 7.5 ppm and a 1H singlet at 8.3 ppm. The pure product gave a molecular ion of 579, consistent with its molecular weight of 578 amu.

Scheme 6



As outlined in Scheme 6, a 500 mL round bottom flask was charged with 10
 5 g (55.84 mmol) of **31**, 84 g (558.4 mmol) of sodium iodide, 20.63 g (55.84 mmol)
 of t-butyl ammonium iodide, and 250 mL acetone. The mixture was stirred at
 reflux overnight. The reaction mixture was filtered to eliminate excess sodium
 iodide, and was diluted with 100 mL hexane. The mixture was filtered again to
 remove more of the remaining sodium iodide. This was repeated until no
 10 precipitate formed when the mixture was diluted with hexane. The reaction yield
 was 9.66 g (77%) of pure 2-(iodomethyl)tetrahydro-2H-pyran **32** as an oil. ^1H

NMR in CDCl_3 was consistent with structure. $R_f=0.55$, using a 2% ethyl acetate/hexane solvent system and a phosphomolybdic acid stain. The product did not give a mass spec signal.

A dry 100 mL round bottom flask (oven heated/argon cooled) was charged with 3.2 g (11.80 mmol) of N-(diphenylmethylene) glycine ethyl ester and purged with argon. Thirty-five milliliters of dry DMPU and 15 mL of dry THF were injected by syringe into the air-free system. The resulting solution was cooled to -78°C , and 17.70 mL (1.5 mmol) of a 0.1 M solution of sodium hexamethylsilazane in THF was injected into the system, which was then stirred at -78°C for 20 minutes. Finally, an air-free solution of 4 g (17.70 mmol) of **32** in dry THF was injected into the system, which was then stirred at -78°C for $\frac{1}{2}$ hr, 0°C for $\frac{1}{2}$ hr, and room temp overnight. The reaction mixture was diluted with 300 mL of ethyl acetate and washed 5 times with 50 mL portions of water to remove the DMPU. The organic layer was extracted and dried over magnesium sulfate. The crude mixture was purified by flash chromatography 4-11% ethyl acetate/hexane solvent system. The yield of the reaction was 1.57 g of the less polar diastereomer of **33**, and 0.33 g of the more polar diastereomer of **33**. The overall yield was 1.9 g (44%). ^1H NMR in CDCl_3 was consistent with structures. The diastereomers have partial overlap by TLC, $R_f=0.55$ using a 5% ethyl acetate/hexane solvent system. The product gave a molecular ion of 366, consistent with its molecular weight of 365.

Note: Throughout the rest of the synthesis, the procedures involve the use of the more polar diastereomer, for the sake of clarity.

The deprotection and work-up of **33** to give **34** follows the same procedure as that for the isonipecotic analogue (see the synthesis of **25** in that sequence).

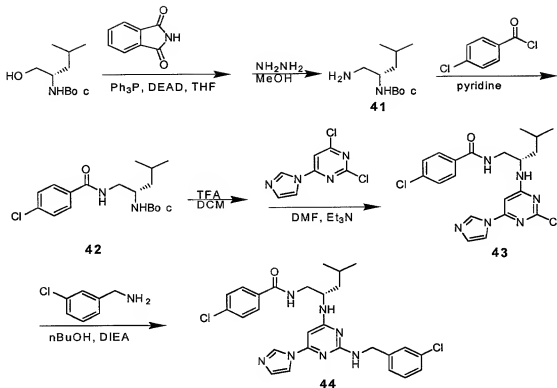
The crude mixture was purified by flash chromatography using an 80-90% ethyl acetate/hexane solvent system and a ninhydrin stain. The yield for the reaction was 68%. ^1H NMR in CDCl_3 was consistent with structures. The $R_f=0.15$

using a 90% ethyl acetate/hexane solvent system. The product gave MH^+ @ 202.

5 The coupling and work-up of **34** with the dichloropyrimidine-pyrrole intermediate follows the same procedure as that for the isonipecotic analogue with the dichloropyrimidine-imidazole intermediate (see the synthesis of **26** in that sequence). The crude mixture was purified by flash chromatography using a 10-20% ethyl acetate/hexane solvent system. The yield for the reaction was 25% for the desired more polar regio-isomer, and 62% for the total yield for both regio-isomers. 1H NMR in $CDCl_3$ was consistent with structure. The $R_f=0.15$ using a 10% ethyl acetate/hexane solvent system. The product gave MH^+ @ 379.

10 The remaining steps from **35** to **36** follow the corresponding procedures as for the isonipecotic analogue (see Scheme 5). The crude **36** was purified by flash chromatography using a 16-25% ethyl acetate/hexane solvent system. 1H NMR in $CDCl_3$ was consistent with structure. The $R_f=0.55$ using a 20% ethyl acetate/hexane solvent system. The product gave MH^+ at 615.

Scheme 7



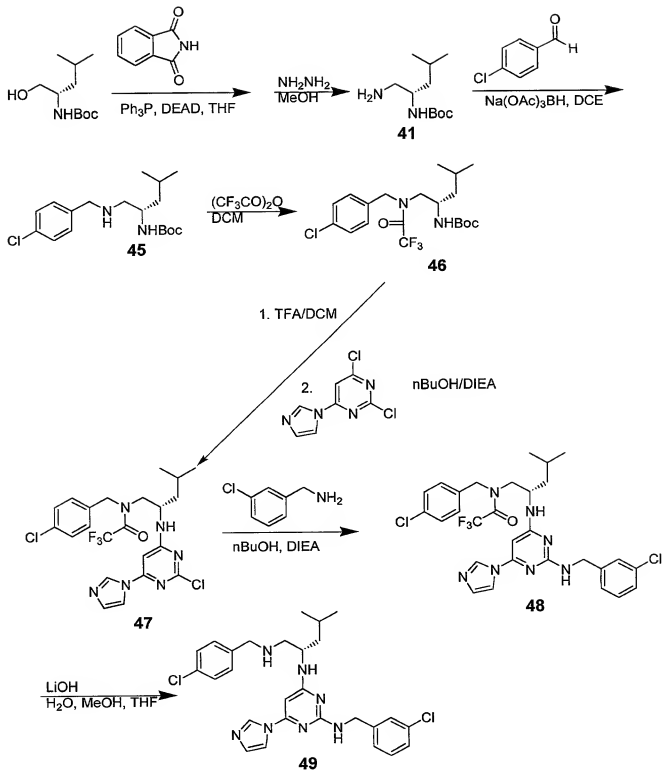
- Scheme 7 depicts an exemplary synthesis wherein $m > \text{zero}$ and $A=A^2$. To
- 5 Boc-D-leucinol (2.7g, 12.4 mmol), triphenylphosphine (3.25g, 12.4 mmol), and phthalimide (1.82g, 12.4 mmol) in 25 mL of dry THF was added DEAD dropwise. The solution was stirred at room temperature overnight, concentrated and taken up in MeOH. To this solution was added hydrazine (780 mL, 24.8 mmol) and heated to reflux for 2 hours. The mixture was allowed to cool to room temperature, and the white precipitate filtered. The mother liquor was concentrated, taken up in
- 10 EtOAc and washed with 1N HCl. The aqueous layer was then cooled in an ice bath, basified with 3N NaOH, and extracted with EtOAc. The organic layer was dried over K₂CO₃ and concentrated to yield **41** as a clear oil. (0.75g, 3.5 mmol, 28%).
- To **41**(0.3g, 1.4 mmol) in 15 mL pyridine was added 4-chlorobenzoyl chloride (194
- 15 mL, 1.5 mmol) and the mixture was stirred at room temperature for 4 hours. The

reaction was poured into 200 mL water and the precipitate filtered. The resulting solid was taken up in DCM and washed with saturated NaHCO_3 and 1M KHSO_4 . The organic layer was dried over MgSO_4 and concentrated to yield **42** as a pale white solid. (0.30g, 0.84 mmol, 61%)

- 5 One hundred sixty-five milligrams of **42** (0.46 mmol) was taken up in 10 mL of DCM and 5 mL TFA was added. After 30 minutes the solution was concentrated, taken up in DMF and basified with excess triethylamine. To this was added 2,4-dichloro-6-imidazolylpyrimidine (100 mg, 0.46 mmol) and the mixture stirred at room temperature overnight. The reaction mixture was concentrated and
- 10 the resulting oil purified on a silica gel column, eluting with 2%MeOH/DCM to yield **43**. (42 mg, 0.1 mmol, 21%).

- To **43** (30 mg, 0.07mmol) in 10 mL n-butanol was added DIEA (60 mL, 0.35 mmol) and 3-chlorobenzylamine (200 mL, 1.4 mmol), and the reaction was heated to 100°C overnight. The solution was concentrated and the resulting oil
- 15 purified on a silica gel column, eluting with 5% MeOH/DCM to yield **44** as a foam. (31 mg 0.06 mmol, 82%).

Scheme 8



Scheme 8 illustrates a similar synthesis to that of Scheme 7 in which A is R^4NH- .

Two hundred seventy milligrams of **41** (1.25 mmol), 4-chlorobenzaldehyde (193 mg, 1.4 mmol) and sodium triacetoxyborohydride (0.4 g, 1.9 mmol) were combined in 20 mL dichloroethane and stirred at room temperature overnight. The mixture was then concentrated, taken up in DCM and washed with saturated $NaHCO_3$, dried over $MgSO_4$ and concentrated to yield **45** which was used without further purification. (0.40g, 1.2 mmol, 94%).

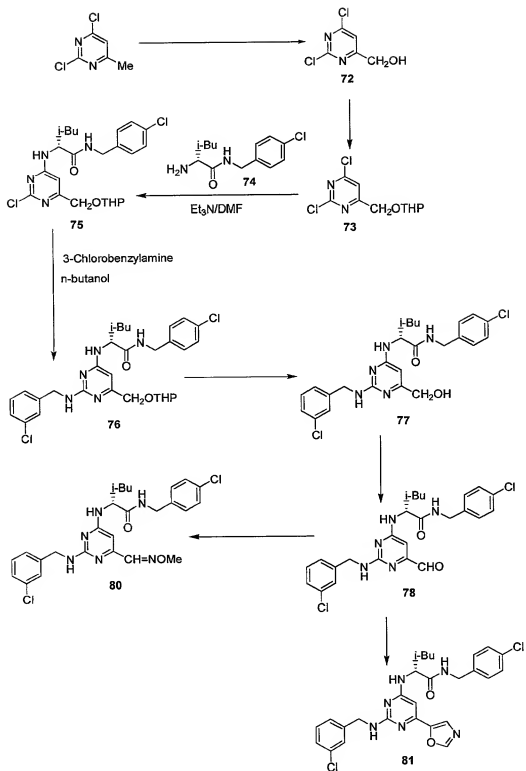
To **45** (0.35 g, 1.03 mmol) in DCM cooled in an ice bath was added trifluoroacetic anhydride (145 μ L, 1.03 mmol) slowly. After 10 minutes the solution was concentrated, taken up in DCM and washed with 1M $KHSO_4$. The organic layer was dried over $MgSO_4$ and concentrated to yield **46** which was used without further purification. (0.32 g, 0.75 mmol, 75%).

Three hundred twenty milligrams of **46** (0.73 mmol) was taken up in 10 mL of DCM and 5 mL TFA was added. After 30 minutes the solution was concentrated, taken up in DMF and basified with excess triethylamine. To this was added 2,4-dichloro-6-imidazolylpyrimidine (190 mg, 0.88 mmol) and stirred at room temperature overnight. The reaction mixture was concentrated and the resulting oil purified on a silica gel column, eluting with 2% MeOH/DCM to yield **47**. (100 mg, 0.19 mmol, 27%).

To **47** (100 mg, 0.19) in 5 mL of n-butanol was added DIEA (60 μ L, 0.35 mmol) and 3-chlorobenzylamine (200 μ L, 1.4 mmol) and heated to 100 °C overnight. The solution was concentrated and the resulting oil purified on a silica gel column, eluting with 5% MeOH/DCM to yield **48** as a foam. (10 mg 0.02 mmol, 9%).

A solution of **48** (10 mg 0.02 mmol) in 10 mL of MeOH:H₂O:THF (1:1:1) was refluxed for 6 hours with excess LiOH. The solution was concentrated, taken up in DCM and washed with brine. The organic layer was dried over MgSO₄ and concentrated to yield **49**. (6 mg, 0.01 mmol, 50%)

Scheme 9



According to Scheme 9, a solution of 2,4-dichloro-6-methylpyrimidine (710 mg, 4.36 mmol) in 4.5 mL of dry THF was added dropwise to a solution of freshly prepared LDA (466 mg, 4.36 mmol) in 17.5 mL of dry THF at -78 °C. After stirring for additional 15 min, the solution of the anion formed was cannulated into a solution of camphorsulfonyloxaziridine (1.0 g, 4.36 mmol) in 11 mL of dry THF maintained at -78 °C. The reaction mixture was stirred in dry ice-acetone bath for 1 h, then quenched with acetic acid and brought to room temperature. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 4:1) gave 300mg of **72**.

A solution of 2,4-dichloro-6-hydroxymethylpyrimidine (1.8 g, 10.0 mmol), dihydropyran (1.26 g, 15 mmol) and PPTS (502 mg, 2.0 mmol) in 20 mL of dry chloroform was stirred for 1 h at RT. TLC indicated complete absence of the starting material. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 85:15) gave 1.02 g of the THP ether **73**.

A solution of leucine amide **74** (254 mg, 1.0 mmol), THP ether (263 mg, 1 mmol) and Et₃N (101 mg, 1 mmol) in 10 mL of dry THF was refluxed for 24 h. Evaporation of the solvent, followed by aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 65:35) gave 174 mg of the desired isomer **75**.

A solution of **75** (174 mg, 0.36 mmol) and 3-chlorobenzylamine (142 mg, 1.0 mmol) in 15 mL of *n*-butanol was refluxed overnight. The solvent was removed *in vacuo* and the residue was purified by chromatography (silica gel, ethyl acetate) to provide 52 mg of **6**.

A solution of **76** (52 mg, 0.085 mmol) and PPTS (50 mg, 0.2 mmol) in 12 mL of 5:1 ethanol:water was refluxed overnight. Evaporation of the solvent and

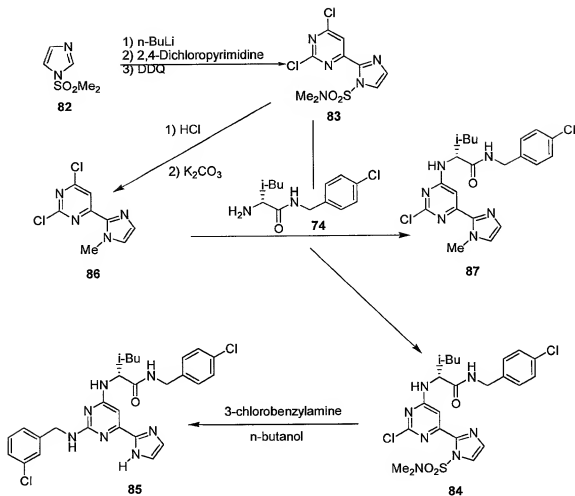
aqueous work up provided 33 mg of alcohol **77**, which was used in the next step without purification.

5 A solution of alcohol **77** (140 mg, 0.28 mmol) and Et₃N (85 mg, 0.84 mmol) in 3 mL of dry DMSO was treated with pyridine.SO₃ complex (134 mg, 0.84 mmol) at RT. Aqueous work up gave the aldehyde **78** in almost quantitative yield.

10 A solution of aldehyde **78** (25 mg, 0.05 mmol), NH₂OMe.HCl (42 mg, 0.5 mmol) and anhydrous NaOAc (41 mg, 0.5 mmol) in 5 mL of ethanol was refluxed overnight. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 3:1) gave 10 mg of oxime ether **80**, (M+H)⁺: 529.2

15 A mixture of aldehyde **78** (135 mg, 0.27 mmol), toluenesulfonylmethyl isocyanide (TOSMIC) (195 mg, 1 mmol) and K₂CO₃ (138 mg, 1 mmol) in 5 mL of methanol was refluxed for 5 h. Evaporation of the solvent and chromatography (silica gel, hexane:ethyl acetate, 1:2) gave 57 mg of oxazole **81**, (M+H)⁺: 539.2.

Scheme 10



- According to Scheme 10, *n*-BuLi (10 mmol, 4 mL of 2.5 M solution in
- 5 hexane) was added at -78 °C to a solution of 1-dimethylsulfamoylimidazole (1.75 g, 10 mmol) in 50 mL of dry ether. After stirring for 1 h, the suspension of the anion formed was quickly transferred by a syringe to a suspension of 2,4-dichloropyrimidine (1.49 g, 10 mmol) in 80 mL of dry ether maintained at -30 °C. After stirring at -30 °C for 30 min, the temperature was brought to 0 °C and
 - 10 maintained there for additional 30 min. The reaction mixture was quenched with a mixture of acetic acid (0.64 mL) water (0.1 mL) and THF (2 mL). Immediately afterwards, a solution of DDQ (2.27 g, 10 mmol) in 10 mL of THF was added and the reaction mixture was stirred overnight. After diluting with ethyl acetate (25

mL), the reaction mixture was filtered through celite and the filtrate was washed with water three times. Finally, a quick wash with ice cold 0.5% NaOH was employed to get rid of the hydroquinone. Evaporation of the solvent and chromatography (silica gel, hexane:ethyl acetate 3:1) provided 550 mg of **83**.

5

A solution of leucine amide **74** (200 mg, 0.79 mmol), 2,4-dichloro-6-(1-dimethylsulfamoylimidazole-2-yl)pyrimidine (254 mg, 0.79 mmol) and Et₃N (88 mg, 0.87 mmol) in 3 mL of DMF was stirred at RT for 5 days. Aqueous work up and chromatography (silica gel, ethyl acetate) gave 200 mg of **84**, (M+H)⁺: 540.1.

10

A solution of chloropyrimidine **84** (200 mg, 0.37 mmol) and 3-chlorobenzylamine (568 mg, 4 mmol) in 10 mL of *n*-butanol was refluxed overnight. The solvent was removed *in vacuo* and the residue was chromatographed (silica gel, ethyl acetate:methanol, 98:2) to provide 12 mg of **85**, (M+H)⁺: 538.2.

15

A solution of 2,4-Dichloro-6-(1-dimethylsulfamoylimidazole-2-yl)pyrimidine **83** (246 mg, 0.76 mmol) in 10 mL of 1.5 N HCl was refluxed for 1 h. After cooling to RT, the pH was adjusted to 8.5 with aq NaHCO₃ and the product was extracted into CH₂Cl₂. After drying the CH₂Cl₂ layer was evaporated to give 110 mg of 2,4-dichloro-6-(imidazole-2-yl)pyrimidine. A mixture of 2,4-dichloro-6-(imidazole-2-yl)pyrimidine (121 mg, 0.56 mmol), K₂CO₃ (100 mg, 0.72 mmol) and CH₃I (2.280 g, 16 mmol) in 15 mL of dry acetone was refluxed for 48 h. After cooling to RT, the solvent was evaporated and the residue was partitioned between water and CH₂Cl₂. The CH₂Cl₂ layer was washed successively with water and brine, and then the solvent was evaporated to give 75 mg of 2,4-dichloro-6-(1-methylimidazole-2-yl)pyrimidine **86**.

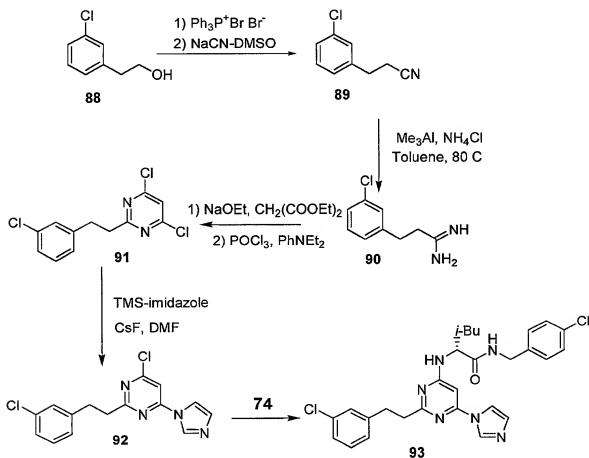
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25

A solution of 2,4-dichloro-6-(1-methylimidazole-2-yl)pyrimidine **86** (72

mg, 0.31 mmol), leucine amide **74** (100 mg, 0.39 mmol) and Et₃N (100 mg, 1 mmol) in 3 mL of DMF was heated to 70 °C overnight. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 1:3) gave 67 mg of **87**, (M+H)⁺: 447.2.

5

Scheme 11



According to Scheme 11, a solution of 3-chlorophenethyl alcohol (5 g, 32 mmol) in 50 mL of dry MeCN was treated with dibromotriphenylphosphane (13.54 g, 32 mmol) for 24 h. The reaction mixture was filtered and the solvent was removed *in vacuo*. The residue was triturated with hexane and filtered. Evaporation of the solvent provided 6.5 g of 3-chlorophenethyl bromide. A solution of the bromide (6.5 g, 29.6 mmol) in 50 mL of dry DMSO containing NaCN (2.17 g, 44 mmol) was heated to 100 °C overnight. The reaction mixture

10

was diluted with water and extracted with ether. The ether layer was washed with water, dried and the solvent was removed *in vacuo*. Chromatography (silica gel, hexane:ethyl acetate, 4:1) provided 3.7 g of nitrile **89**.

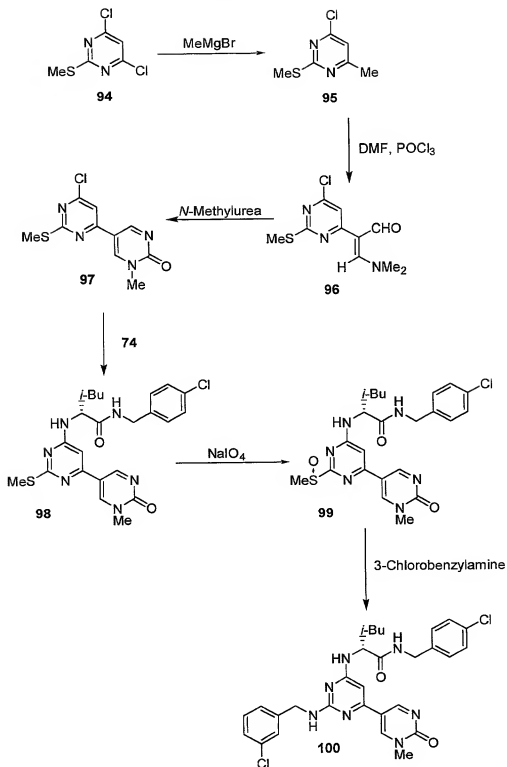
5 A 2 M solution of Me_3Al in toluene (18 mL, 36 mmol) was slowly added to
a stirred suspension of NH_4Cl (2.07 g, 38.7 mmol) in 20 mL of dry toluene at 5 °C.
6 After the addition was over, the reaction mixture was warmed to RT and stirred
for 2 h. Then, a solution of nitrile **89** (3.7 g, 22.4 mmol) in 15 mL of dry toluene
was added and the solution was heated to 80 °C for 18 h. After cooling to RT, the
reaction mixture was poured into a slurry of 15 g of silica gel in 50 mL of CHCl_3
10 and stirred for 5 min. The silica gel was filtered and washed with methanol. The
filtrate and washings were combined and the solvent was removed. The residue
obtained was partitioned between water and methylene chloride. Evaporation of
the methylene chloride provided 2.7 g of amidine **90**.

A solution of amidine **90** (2.7 g, 14.8 mmol) and diethyl malonate (2.37 g,
15 14.8 mmol) in 50 mL of dry ethanol containing freshly prepared NaOEt (1.0 g,
14.8 mmol) was refluxed for 15 h. After cooling to RT, the solvent was removed
and the residue was dissolved in water. The pH was adjusted to 4 and the
precipitated solid was filtered and dried to provide 2.6 g of 2-(3-chlorophenethyl)-
4,6-dihydroxy-pyrimidine. A mixture of 2-(3-chlorophenethyl)-4,6-
20 dihydroxypyrimidine (2.6 g, 10.38 mmol), POCl_3 (25 mL) and *N,N*-diethylaniline
(6 mL) was refluxed overnight. After cooling to RT, the reaction mixture was
poured into ice water and the product was extracted into ether. The ether layer was
washed successively with water and brine and the solvent was evaporated.
Chromatography (silica gel, hexane:ethyl acetate, 9:1) of the oil provided 2.6 g of
25 the 2-(3-chlorophenethyl)-4,6-dichloropyrimidine (**91**).

A solution of 2-(3-chlorophenethyl)-4,6-dichloropyrimidine (286 mg, 1 mmol) **91** in 3 mL of dry DMF was treated with 1-trimethylsilylimidazole (140 mg, 1 mmol) and CsF (152 mg, 1 mmol) at RT overnight. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 1:1) gave 200 mg of 4-chloro-2-(3-chlorophenethyl)-6-(1-imidazolyl)pyrimidine (**92**).

A solution of 4-chloro-2-(3-chlorophenethyl)-6-(1-imidazolyl)pyrimidine **92** (100 mg, 0.31 mmol), leucine amide **74** (95 mg, 0.372 mmol) and DIEA (129 mg, 1 mmol) in 2 mL of DMF was heated to 80 °C for 24 h. Aqueous work up and chromatography (silica gel, ethyl acetate:methanol, 98:2) gave 105 mg of **93**,
(M+H)⁺: 537.4

Scheme 12



According to Scheme 12, a solution of 4,6-dichloro-2-methylthiopyrimidine (1.95 g, 10 mmol) in 30 mL of dry THF was cooled to 0 °C and treated with a solution of MeMgBr (14 mL of 1.4 M solution, 19.6 mmol). After overnight stirring at RT, the reaction mixture was quenched with sat. NH₄Cl. The organic layer washed with brine, dried and evaporated. The residue was purified by chromatography (silica gel, hexane:ethyl acetate, 9:1) to provide 1.3 g of 4-chloro-6-methyl-2-methylthiopyrimidine (**95**).

Dry DMF (2 mL) was cooled to -5 °C and POCl₃ (15.4 mmol, 2.31 g) was added dropwise. The cooling bath was removed and the reaction mixture was stirred for 15 min at RT. 4-chloro-6-methyl-2-methylthiopyrimidine (1.3 g, 7.47 mmol) was added and the contents were heated to 60 °C overnight. The reaction mixture was poured on ice, pH was adjusted to 9 and the precipitated product was filtered. The precipitate was washed with water and dried to provide 1.3 g of the enaminone **96**.

A mixture of enaminone **96** (675 mg, 2.6 mmol) and *N*-methylurea (232 mg, 3.14 mmol) in 5 mL of acetic acid was heated to 100 °C for 2 h. Aqueous work up and chromatography (silica gel, ethyl acetate:methanol, 98:2) gave 100 mg of pyrimidinone **97**.

A solution of **97** (100 mg, 0.37 mmol), leucine amide **74** (100 mg, 0.34 mmol) and DIEA (60 mg, 0.46 mmol) in 3 mL of DMF was heated to 80 °C for 2 days. Aqueous work up followed by chromatography (silica gel, ethyl acetate:methanol, 95:5) gave 30 mg of **98**.

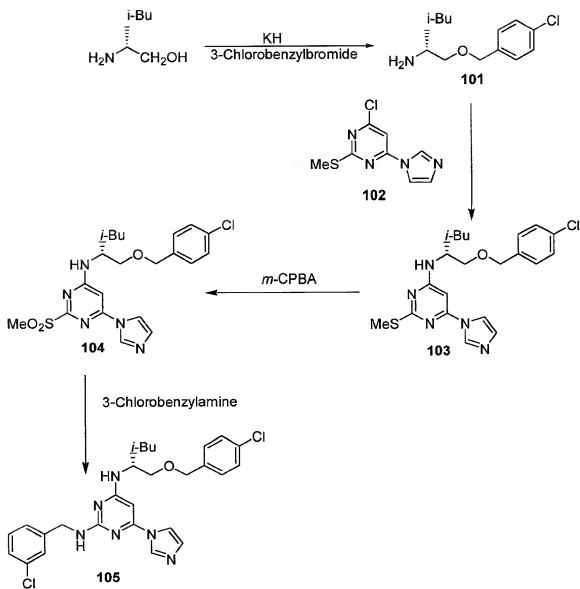
A mixture of **98** (30 mg, 0.061 mmol) and NaIO₄ (263 mg, 1.23 mmol) in 6 mL of 1:1 methanol:water was stirred overnight at RT. Aqueous work up gave 10

mg of the crude sulfoxide **99**.

The sulfoxide **99** (10 mg, 0.002 mmol) and 3-chlorobenzylamine (27 mg, 0.2 mmol) in 2 mL of *n*-butanol were heated to reflux for 24 h. Aqueous work up after removal of *n*-butanol, followed by chromatography (silica gel,

- 5 CH₂Cl₂:methanol, 95:5) gave 2 mg of **100**, (M+H)⁺: 580.2.

Scheme 13



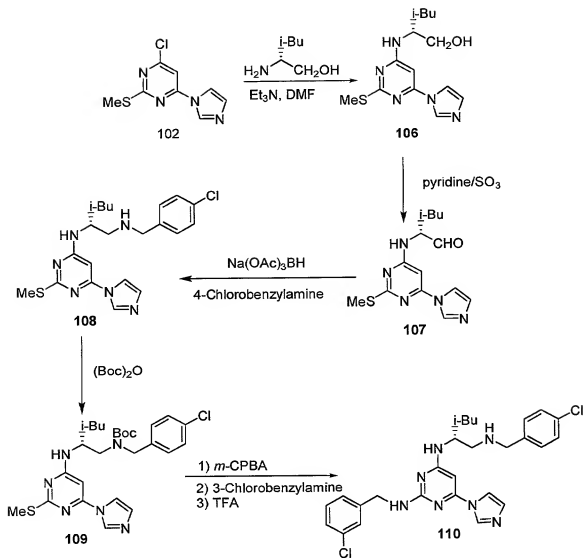
According to Scheme 13, a solution of (*R*)-leucinol (1.288 g, 11 mmol) in 5 mL of THF at RT was added dropwise to a stirred suspension of potassium hydride (0.485 g, 12.1 mmol) in 25 mL of dry THF. After overnight stirring at RT, a solution of 4-chlorobenzylbromide (2.25g, 11 mmol) in 5 mL of THF was added dropwise. The stirring was continued for additional 3 h. The solvent was evaporated and the residue was partitioned between water and ether. The ether layer was washed with brine, dried and the solvent was removed *in vacuo* to provide 2.1 g of ether **101**.

A solution of 4-chloro-6-(1-imidazolyl)-2-methylthiopyrimidine (227 mg, 1 mmol), aminoether **101** (242 mg, 1 mmol) and Et₃N (101 mg, 1 mmol) in 4 mL of DMF was heated to 70 °C for 24 h. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 1:1) provided 300 mg of thioether **103**.

A solution of the thioether **103** (300 mg, 0.7 mmol) in 10 mL of CH₂Cl₂ was treated with *m*-CPBA (428 mg, 1.74 mmol) at 0 °C overnight. The precipitate was filtered and the filtrate was evaporated to obtain crude sulfone **104**. No starting material or intermediate sulfoxide was detected by MS.

A solution of sulfone **104** (100 mg, 0.22 mmol) and 3-chlorobenzylamine (2 mmol) in 3 mL of *n*-butanol was refluxed for 24 h. Aqueous work up after removal of *n*-butanol, followed by chromatography (silica gel, ethyl acetate) gave 22 mg of **105**, (M+H)⁺: 525.2.

Scheme 14



According to Scheme 14, a solution of 4-chloro-6-(1-imidazolyl)-2-methylthiopyrimidine (227 mg, 1 mmol), (*R*)-leucinol (117 mg, 1 mmol) and Et₃N (101 mg, 1 mmol) in 3 mL of DMF was heated to 70 °C for 24 h. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 1:3) gave 290 mg of **106**.

5

A solution of alcohol **106** (290 mg, 0.94 mmol) and Et₃N (303 mg, 3 mmol) in 5 mL of DMSO was treated with pyridine-sulfur trioxide complex (477 mg, 3 mmol) at RT overnight. Aqueous work up gave 280 mg of the crude aldehyde **107** which was used in the next step without purification.

10

A mixture of aldehyde **107** (280 mg, 0.91 mmol), Na(OAc)₃BH (290 mg, 1.37 mmol), 4-chlorobenzylamine (142 mg, 1 mmol) and HOAc (60 mg, 1 mmol) in 10 mL of dry 1,2-dichloroethane was stirred at RT overnight. Aqueous work up and chromatography (silica gel, CH₂Cl₂:methanol:NH₄OH, 95:5:0.5) gave 135 mg of **108**.

15

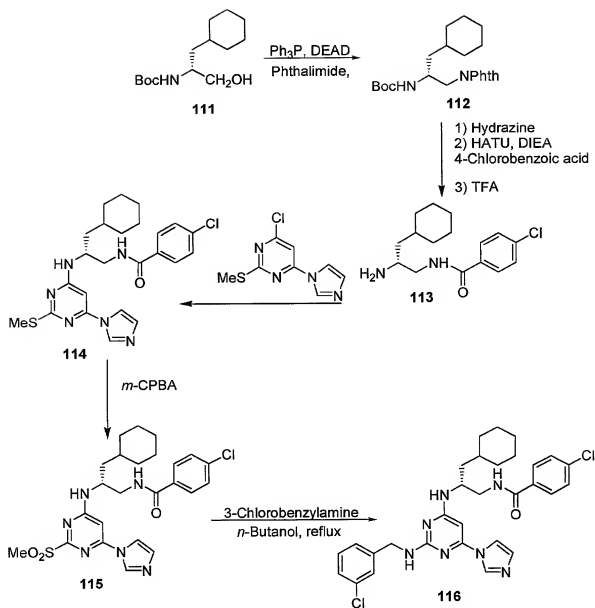
A solution of amine **108** (130 mg, 0.3 mmol) and boc-anhydride (214 mg, 1 mmol) in 5 mL of THF was stirred at RT overnight. Aqueous work up after removal of the solvent, provided 60 mg of the Boc-protected amine **109**.

20

A mixture of the Boc-protected amine **109** (60 mg, 0.11 mmol) and *m*-CPBA (83 mg, 0.33 mmol) in 20 mL of 1:1 CH₂Cl₂:phosphate buffer was stirred at 0° C for 2 h and then kept in the refrigerator overnight. The methylene chloride layer was filtered and the solvent was removed to provide the crude sulfone. A solution of the sulfone in 5 mL of *n*-butanol containing 10 eq of 3-chlorobenzylamine was refluxed for 20 h. The solvent was removed *in vacuo* and the residue was treated with 2:1 CH₂Cl₂:TFA for two days. After removal of the solvent, the residue was taken in water and basified. The precipitated product was extracted into CH₂Cl₂. Evaporation of the CH₂Cl₂ layer gave 6 mg of **110**, (M+H)⁺: 524.2.

25

Scheme 15



According to Scheme 15, a solution of Ph₃P (262 mg, 1 mmol) and phthalimide (147 mg, 1 mmol) in 3 mL of dry THF was treated with a solution of diethyl azodicarboxylate (174 mg, 1 mmol) in 2 mL of dry THF at RT. After stirring for 5 min, a solution of alcohol **111** (257 mg, 1 mmol) in 5 mL of dry THF was added and the stirring was continued for 3 days. The solvent was removed and the residue was chromatographed (silica gel, hexane:ethyl acetate, 4:1) to obtain 320 mg of phthalimide **112**.

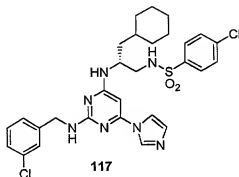
Three hundred twenty milligrams (0.83 mmol) of phthalimide **112** and 50 mg (1 mmol) of NH₂NH₂·H₂O in 5 mL of ethanol was refluxed for 2 h. The solvent was removed and the residue was partitioned between CH₂Cl₂ and 1 N NaOH. Evaporation of the organic layer after drying provided the primary amine. The amine was coupled with 4-chlorobenzoic acid (130 mg, 0.83 mmol) using HATU (1 eq) in DMF containing 2 eq of DIEA. The amide was purified by chromatography (silica gel, hexane:ethyl acetate, 1:1), yield 200 mg. The boc group was removed by stirring in TFA:CH₂Cl₂ (1:2) at RT overnight.

A solution of 4-chloro-6-(1-imidazolyl)-2-methylthiopyrimidine (227 mg, 1 mmol), TFA salt of amine **113** (220 mg, 1 mmol) and Et₃N (303 mg, 3 mmol) in 3 mL of DMF was heated to 80 °C overnight. Aqueous work up and chromatography (silica gel, hexane:ethyl acetate, 1:3) gave 130 mg of **114**.

A solution of the thioether **114** (130 mg, 0.27 mmol) in 20 mL of CH₂Cl₂ was treated with *m*-CPBA (196 mg, 0.8 mmol) at 0 °C for 1 h, and then left in a refrigerator overnight. The reaction mixture was filtered and the crude sulfone **115** was isolated by evaporation of the filtrate.

A solution of sulfone **115** (130 mg, 0.25 mmol), 3-chlorobenzylamine (72

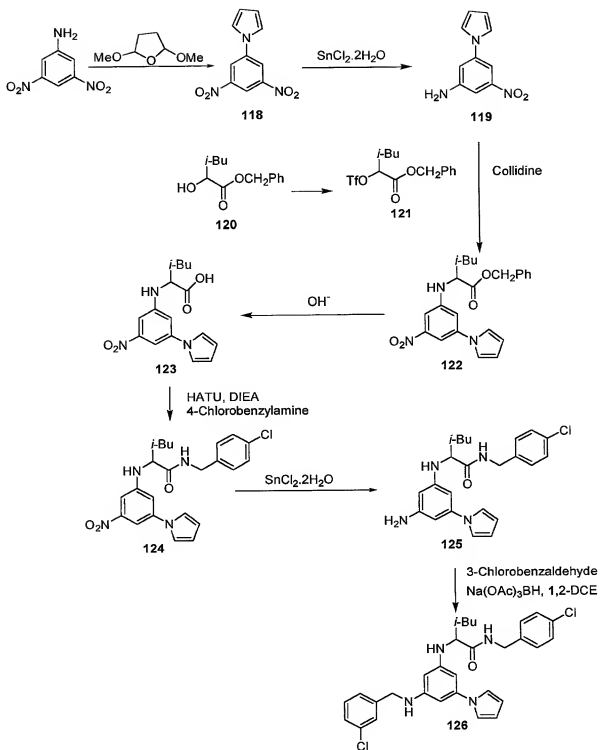
mg, 0.5 mmol) and Et₃N (50 mg, 0.5 mmol) in 4 mL of *n*-butanol was heated to reflux for 20 h. Aqueous work up and chromatography (silica gel, ethyl acetate:methanol, 99:1) gave 66 mg of **116**, (M+H)⁺: 578.2.



- 5 The corresponding sulfonamide **117** was prepared by a similar procedure to that of Scheme 15, using 4-chlorobenzenesulfonyl chloride in place of 4-chlorobenzoyl chloride, (M+H)⁺: 614.2.

Compounds in which X, Y and Z are CH and Q is pyrrole are prepared as shown in Scheme 16

Scheme 16



According to Scheme 16, a solution of 3,5-dinitroaniline (1.83 g, 10 mmol) and 2,5-dimethoxytetrahydrofuran in 20 mL of HOAc was refluxed overnight. The reaction mixture was poured into water and extracted with EtOAc. The ethyl acetate layer was washed with water followed by aq NaHCO₃ and brine. After drying, the solvent was removed to provide 1.52 g of 1-(3,5-dinitrophenyl)pyrrole.

A mixture of 1-(3,5-dinitrophenyl)pyrrole (1.52 g, 6.52 mmol) and SnCl₂·2H₂O (4.4 g, 19.57 mmol) in 30 mL of ethyl acetate was stirred over weekend at RT. The solvent was removed and the residue was taken in water. The aqueous layer was basified with 1 N NaOH to dissolve the tin salts, and the product was extracted into ethyl acetate. Chromatography (silica gel, hexane:ethyl acetate, 4:1) of the crude product provided 440 mg of 1-(3-amino-5-nitrophenyl)pyrrole.

A solution of benzyl ester **120** (222 mg, 1 mmol), DIEA (129 mg, 1 mmol) and triflic anhydride (282 mg, 1 mmol) in 5 mL of dry CH₂Cl₂ was stirred at 0 °C for 1.5 h. TLC in hexane:ethyl acetate (4:1) indicated complete conversion of the starting material. The solvent was removed and the crude triflate **121** was used for the next step.

A solution of 1-(3-amino-5-nitrophenyl)pyrrole (203 mg, 1 mmol) and triflate **121** in 15 mL of 1,2-dichloroethane containing collidine (121 mg, 1 mmol) was refluxed for 24 h. Aqueous acidic work up, followed by chromatography (hexane:ethyl acetate, 4:1 gave 95 mg of **122**.

The ester **122** (95 mg, 0.23 mmol) was treated 250 mg of NaOH in 5 mL of 95:5 methanol:water. After overnight stirring at RT, the solvent was removed and the residue was taken in water. The pH was adjusted to 3 and the precipitated acid was extracted into ethyl acetate. Evaporation of the ethyl acetate layer gave 57 mg

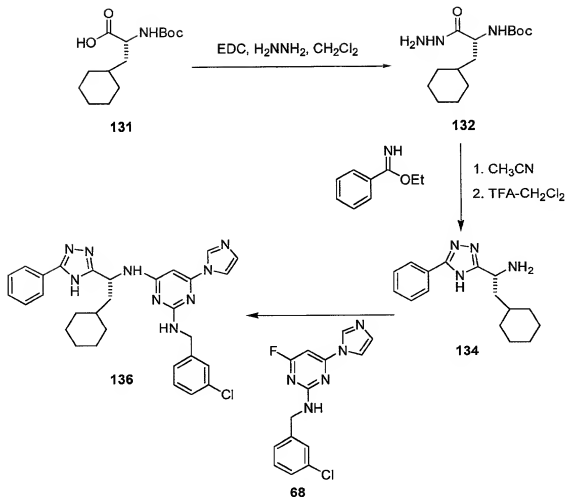
of **123**.

To a solution of carboxylic acid (57 mg, 0.18 mmol) in 3 mL of dry DMF containing 2 eq of DIEA, 1 eq of HATU was added. After 5 min 1 eq of 4-chlorobenzyl amine was added and the stirring was continued overnight. The crude
5 product **124** obtained after aqueous work up was used directly for the next step.

Amide **124** was reduced with $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$ (5 eq) in ethyl acetate as described earlier. The aniline **125** was purified by chromatography (silica gel, hexane:ethyl acetate, 1:1), yield 7 mg.

A mixture of aniline **125** (7 mg, 0.017 mmol), $\text{Na}(\text{Oac})_3\text{BH}$ (6 eq) and 3-
10 chlorobenzaldehyde (6 eq) in 2 mL of 1,2-dichloroethane was stirred at RT overnight. Aqueous work up and chromatography (silica gel, hexane: ethyl acetate, 62:38) gave 3 mg **126**, $(\text{M}+\text{H})^+$: 535.1.

Scheme 17



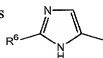
To a solution of N-BOC-cyclohexyl alanine (200 mg, 0.74 mmol) in 2 mL of dry methylenechloride was added dropwise hydrazine (0.1 mL, 0.89 mmol) and EDC (159 mg, 0.81 mmol) at 23 °C. The reaction mixture was stirred for 48 h, then washed with NH_4Cl , water, and brine to give 150 mg of **132**.

A solution of hydrazide **132** (72.4 mg, 0.254 mmol) and imidate **133** (52 mg, 0.28 mmol) in 2 mL of dry acetonitrile was stirred for 16 h at RT. TLC indicated complete absence of the starting material. Solvent was removed and the crude product was treated with TFA:methylenechloride, 1:1, and washed with 1 N

NaOH, water and brine to give 16.2 mg of the triazole **134**.

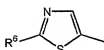
A solution of triazole **134** (16.2 mg, 0.06 mmol), fluoropyrimidine **68** (27.3 mg, 0.09 mmol) and $i\text{Pr}_2\text{NEt}$ (0.02 mL, 0.12 mmol) in 1 mL of dry $n\text{BuOH}$ was refluxed for 16 h. Evaporation of the solvent, followed by aqueous work up and chromatography (silica gel, hexane:ethyl acetate:methanol, 4:4:1) gave 9.0 mg of the desired product **136**.

Compounds of the invention in which A^1 is



and in which A^1

is

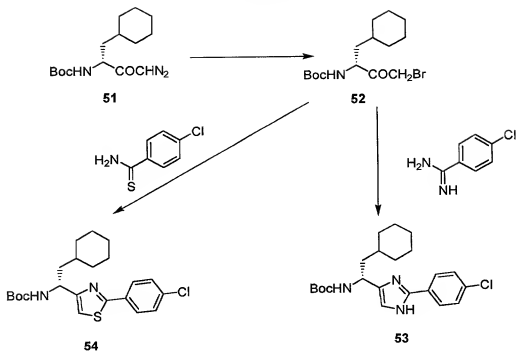


are synthesized as shown in Scheme 17. In both cases the Boc

protecting group is cleaved with trifluoroacetic acid and the amine is reacted

further as already described.

Scheme 18



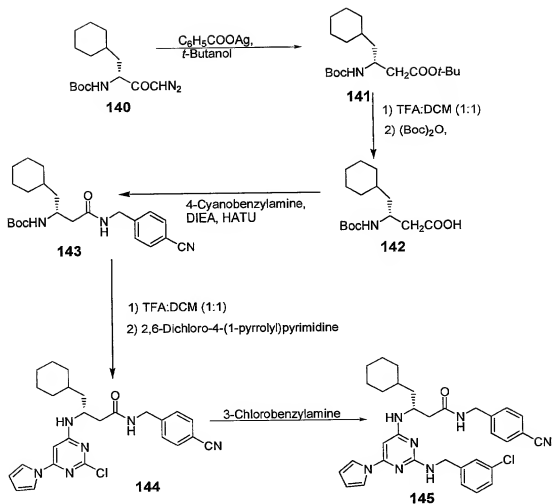
A solution of diazo ketone **51** (2.89 g, 9.78 mol.) in 60 mL of ether was cooled to -20 °C and 2 mL of 48% HBr (960 mg, 11.85 mol.) was added dropwise.

After 35 minutes, an additional 0.5 mL of HBr (240 mg, 2.96 mol.) was added and the stirring was further continued for 25 min. TLC [hexane:ethyl acetate (4:1)] indicated complete absence of the starting material and appearance of the less polar α -bromoketone. Cold aqueous work-up and chromatography on silica gel with hexane:ethyl acetate (85:15) gave 2.7 g of the pure α -bromoketone **52**. ¹H NMR

(CDCl₃): 5.00-4.80 (m, 1H), 4.64-4.50 (m, 1H), 1.90-0.90 (m, 22H). The α -bromoketone is reacted with 4-chlorobenzamidine in refluxing chloroform to provide the imidazole **53** according to the method of Nagao et al. [*Heterocycles* **42**, 517-523 (1996)]. The α -bromoketone is reacted with 4-chlorothiobenzamidine in dioxane to provide the thiazole **54** according to the method of Nan'Ya et al. [*J.*

Heterocycl. Chem. **32**, 1299-1302 1995].

Scheme 19



water. The ether layer was successively washed with aq NaHCO₃, water and brine. After drying (MgSO₄), the ether was evaporated to give the diazoketone **140** as a pale yellow oil.

The diazoketone was dissolved in 10 mL of *t*-butanol and the solution was brought to reflux under argon. A freshly prepared and filtered solution of silver benzoate (0.5 g, 2.18 mmol) in 3 mL of Et₃N was added dropwise over 30 min *via* syringe. The reflux was continued for an additional 1 h. A small amount of decolorizing carbon was added and the reaction mixture was filtered through celite. After evaporation of the filtrate, the residue was chromatographed (silica, hexane:ethyl acetate (85:15)) to give 650 mg of *R-t*-butyl 3-(cyclohexylmethyl)-3-*t*-butoxycarbonylamino propionate, **141** (M+H)⁺: 342.0.

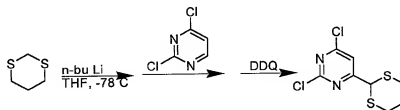
A solution of **141** (650 mg, 1.90 mmol) in 10 mL of TFA:DCM (1:1) was stirred for 6 h at RT. The solvent was removed and the residue was treated with Boc-anhydride in dioxan-aq NaOH to give 486 mg of *R*-3-(cyclohexylmethyl)-3-*t*-butoxy carbonylamino-propionic acid, **142** (M-H)⁺: 284.7

A solution of **142** (284 mg, 1.0 mmol) and DIEA (258 mg, 2.0 mmol) in 5 mL of dry DMF was treated with HATU (380 mg, 1.0 mmol) at RT. After 5 min, 4-cyanobenzylamine (132 mg, 1.0 mmol) was added and the reaction mixture was stirred overnight at RT. Aqueous workup and chromatography (silica gel, hexane:ethyl acetate (1:3)) gave 200 mg of the amide **143**.

A solution of the amide (200 mg, 0.5 mmol) in 10 mL of TFA:DCM (1:1) was stirred at RT for two days. The solvent was evaporated and the residue was taken in 5 mL of DMF containing DIEA (258 mg, 2.0 mmol) and 2,6-dichloro-4-(1-pyrrolyl)pyrimidine (107 mg, 0.5 mmol). After heating overnight at 80 °C, the reaction mixture was diluted with water and the product was extracted into ethyl acetate. The solvent was removed and the residue was chromatographed (silica gel, hexane:ethyl acetate (1:3)) to give 50 mg of the 2-(1-pyrrolyl)pyrimidine derivative and 58 mg of the 4-(1-pyrrolyl)pyrimidine compound **144**, (M+H)⁺: 477.3

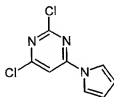
A solution of **144**, (30 mg, 0.063 mmol) and 3-chlorobenzylamine (50 mg, 0.35 mmol) in 2 mL of *n*-butanol was refluxed overnight. The solvent was removed and the residue was purified by chromatography (silica gel, hexane:ethyl acetate (1:3)) to give 4 mg of **145**, (M+H)⁺: 582.3

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To 1,3-dithiane (6.2g, 50.0 mmol) in 20 mL dry THF was added *n*-butyl lithium (2.5M, 22mL, 55.0 mmol) dropwise while cooling to -78°C. After 30 minutes a solution of 2,4-dichloropyrimidine (10.0g, 75 mmol) in 15 mL dry THF was added dropwise. After 30 minutes the mixture was warmed to 0° and DDQ (12.5g, 55.0 mmol) was added and allowed to warm to room temperature. After 1 hour the mixture was concentrated and the resulting residue purified on a silica gel column, eluting with 3:7 EtOAc:hexanes to yield 2,4-dichloro-6-(2-dithianyl)pyrimidine as a light yellow oil (1.2g, 5.5 mmol, 9%)

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2,6-Dichloro-4-(1-pyrrolyl)pyrimidine was prepared as follows: A dry 500 mL round bottom flask (oven-heated/argon cooled), was charged with 2.97 g (74.34 mmol) of a 60% dispersion of sodium hydride in mineral oil. The flask was purged with argon, and 200 mL of hexane were quickly added. The mixture was purged again, and stirred for 5-10 minutes. The stirring was then stopped, and the

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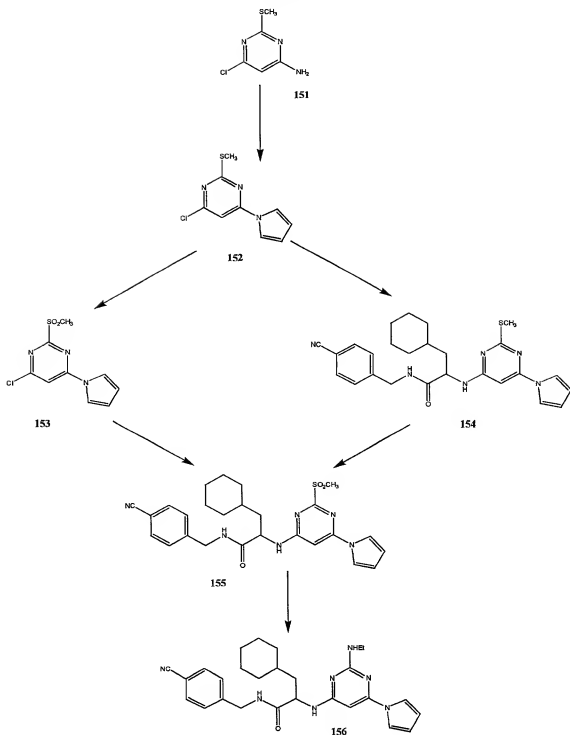
sodium hydride was allowed to settle, at which point the hexane was quickly decanted off. The mixture was purged with argon again and the rinsing was repeated, to ensure the reaction is free from the mineral oil suspension. Next, 200 mL of dry THF were injected by syringe into the air-free mixture. The mixture was then cooled to 0°C, and connected to an oil-bubbler. Then 3.44 mL (49.60 mmol) of pyrrole were injected into the mixture by syringe (vigorous bubbling occurred as hydrogen evolved), and it was stirred for 1 hr. Finally, 10 g (54.52 mmol) of 2,4,6-trichloropyrimidine were injected quickly into the reaction mixture, and it was vigorously stirred overnight. The reaction mixture was diluted with 200 mL of ethyl acetate and washed with a solution of 14.5 g (75 mmol), of citric acid in 100 mL of water. The organic layer was extracted and dried with magnesium sulfate. The mixture was then concentrated down to give a brown, viscous material. The crude material was loaded relatively quickly onto a chromatographic column (25'' x 3''), which was filled with 11 ¼'' silica gel. Elution was started at 40:1 hexane/ether for about 2 L, and then the concentration was increased to 35:1 hexane/ether for about 4 L. The best TLC system was 9:1 hexane/ether. With that system, the four product spots could be seen: the top spot was the regio-isomer with the pyrrole substituted on the 2-position of the pyrimidine, the second spot was unreacted pyrimidine, the third spot was the regio-isomer with the pyrrole substituted at the 4-position (desired product), and the most polar spot was a bis-addition product. Most of the desired product was separated with the column (2.5 g), but the remaining mixture with the bis-product was recrystallized from hexane to give another 1.5 g. The total yield was 4 g (38%) of the white solid. ¹H NMR in CDCl₃: a 2H triplet at 6.42 ppm (J=2.55 Hz), a 1H singlet at 7.16 ppm, and a 2H triplet at 7.48 ppm (J=2.55 Hz). In 9:1 hexane/ether, the R_f = 0.37. This compound did not give a mass spec signal.

The corresponding 2,6-difluoro-4-(1-pyrrolyl)pyrimidine is made in analogous fashion from 2,4,6-trifluoropyrimidine. Both are useful as intermediates

in the synthesis of B₁-BK antagonists of the invention. An improved synthesis of 2,6-dichloro-4-(1-pyrrolyl)pyrimidine proceeds from 4-amino-2,6-dichloropyrimidine. A mixture of 4-amino-2,6-dichloropyrimidine (5.0 g, 30.5 mmol) and 2,5-dimethoxytetrahydrofuran (4.03 g, 30.5 mmol) in 100 mL of HOAc was refluxed for 2 hours. The reaction mixture was cooled to RT and poured into large quantity of water. The crude product was extracted into ethyl acetate and the ethyl acetate layer was extracted successively with water, aqueous NaHCO₃ and brine. The organic layer was dried (MgSO₄) and the solvent was evaporated. The residue was purified by chromatography (silica gel, hexane:ethyl acetate (96:4) to provide 4.4 g (73%) of 2,6-dichloro-4-(1-pyrrolyl) pyrimidine. ¹H NMR (CDCl₃): δ (ppm) 6.4 (s,2H), 7.15 (s, 1H), 7.5 (s, 2H).

As described above, both the dichloro and the difluoro-intermediates provide mixtures of regioisomers when reacted with nucleophiles (cf. **144** in Scheme 19). Although this is useful when both regioisomers are desired, the route shown in Scheme 20 below provides a regioselective synthesis. According to Scheme 20, 4-amino-6-chloro-2-methylthiopyrimidine **151** was reacted with 1 equivalent of 2,5-dimethoxytetrahydrofuran in refluxing acetic acid to provide 6-chloro-2-methylthio-4-(1-pyrrolyl)pyrimidine **152**: ¹H NMR (CDCl₃) δ 2.75 (s,3H), 6.55 (d,2H), 7.05 (s,1H), 7.65 (d,2H). The 6-chloro-2-methylthio-4-(1-pyrrolyl)pyrimidine **152** is either (a) oxidized with 2.2 equivalents of m-chloroperoxybenzoic acid in dichloromethane at 0 °C to provide 6-chloro-2-methylsulfonyl-4-(1-pyrrolyl)pyrimidine **153** or (b) reacted with 1 equivalent of the N-(p-cyanobenzyl)amide of cyclohexylalanine and 1 equivalent of diisopropylethylamine in DMF at 80 °C to provide the 2-methylthiopyrimidine **154**. The oxidation and nucleophilic displacement steps are then reversed [i.e. **153** is reacted according to (b) or **154** is reacted according to (a)] to provide the 2-methylsulfonylpyrimidine **155**, which is dissolved in n-butanol saturated with ethylamine and heated in a sealed tube to produce **156**.

Scheme 20



Bioassays

- Tissues are taken from New Zealand white rabbits (1.5-2.5 kg) and Duncan Hartley guinea pigs (250-350g) of either sex, killed by stunning and exsanguination. Human umbilical cords are obtained after spontaneous delivery at term. The rabbit jugular vein (RbJV) and the guinea pig ileum (GPI), are two preparations containing B₂ receptors. The rabbit aorta (RbA) contains B₁ receptors, and the human umbilical vein (HUV) is a mixed preparation containing both B₁ and B₂ receptors. Helical strips of RbJV, treated with 1 µmol/L of captopril to avoid peptide degradation, are prepared according to Gaudreau et al. [Can. J. Physical. Pharmacol. **59**, 371-379 (1981)] Helical strips of RbA devoid of endothelium are prepared according to Furchgott and Bhadrakom. [J. Pharmacol. Exp. Ther. **108**, 124-143 (1953)] Longitudinal segments of GPI are prepared with the procedure described by Rang [Brit. J. Pharmacol. **22**, 356-365 (1964)]. Helical strips of HUV are prepared according to Gobeil et al. [Brit. J. Pharmacol. **118**, 289-294 (1996)]. Unless otherwise indicated below, the tissues are suspended in 10-mL organ baths containing warm (37°C), oxygenated (95% O₂- 5% CO₂) Krebs solution of the following composition in mmol/L; NaCl: 118. 1; KCl: 4.7; CaCl₂·6H₂O: 2.5; KH₂PO₄: 1.2; MgSO₄·7H₂O: 1.18; NaHCO₃: 25.0 and D-Glucose: 5.5. The RbA are stretched with an initial tension of 2 g, whereas the RbJV and the GPI are loaded with 0.5 g. Changes of tension produced by the various agents are measured with Grass isometric transducers (model FT 03C, Grass Instrument Co., Quincy, Mass.). Myotropic contractions are displayed on a polygraph. Before testing the drugs, the tissues are allowed to equilibrate for 60-120 minutes, during which time the tissues are repeatedly washed and the tension readjusted every 15 min.

At the beginning of each experiment, a submaximal dose of bradykinin

- (BK) (9 nmol/L), is applied repeatedly on the RbJV, the GPI or the HUV to ensure that tissues responded with stable contractions. In the RbA, the B₁ preparation whose response has been shown to increase during the incubation *in vitro*, desArg⁹ K (550 nmol/L) are applied 1,3 and 6 h after the equilibration period, in order to
- 5 monitor the progressive increase of sensitivity of the tissue which generally reaches the maximum after 3-6 h.

- Repeated applications of a single and double concentration of BK (on RbJV, GPI and HUV) and of desArg⁹BK (RbA and HUV) are made in the absence and in presence of the test compounds to evaluate their apparent affinities as
- 10 antagonists, in terms of pA₂ (-log₁₀ of the molar concentration of antagonist that reduces the effect of a double concentration of agonist to that of a single one). The antagonists are applied 10 min before measuring the myotropic effects of either BK (the B₂ receptor agonist) or desArg⁹BK (the B₁ receptor agonist). Pharmacological assays on the HUV (a mixed B₁ and B₂ receptor preparation) are done in presence
- 15 of either HOE140 (400 nmol/L) (a potent B₂ receptor antagonist) or Lys[Leu⁸]des Arg⁹BK (1 μmol/L) (a potent B₁ receptor antagonist) (applied 10 min prior to the tested agents) to study the B₁ and the B₂ receptors, respectively. All kinin antagonists are initially applied to tissues at concentration of 10 μg/mL to measure their potential agonistic activities (α^B) in comparison with BK (in the B₂ receptor
- 20 preparations) or desArg⁹BK (in the B₁ receptor preparations). The compounds of the present invention exhibit inhibition at very low concentrations only when tested in human or primate systems; thus the foregoing (and following) tests in rabbit and rodent tissues are useful only for demonstrating lack of undesired effects on other receptors than B₁ and in other tissues than human. In order to determine the
- 25 potency of compounds of the invention, those tests that employ rabbit and rodent tissues are modified to employ human and primate tissues, as well known to persons in the art.

Streptozotocin has been extensively used to produce type I diabetes in animals. This experimental disease is characterized by a mild inflammatory reaction in the Langerhans islets. Male C57L/K₃ mdb mice are injected with streptozotocin (40mg/kg) for 5 consecutive days. The kinin B₁ receptor antagonists are injected subcutaneously to STZ mice at 300 µg/Kg bw twice a day and 500 µg/Kg per day, respectively. Treatment with antagonists is started 3 days after STZ and lasts for 10 days. Plasma glucose is determined by the glucose oxidase method, and urinary samples are assayed at 13 days for proteins, nitrites and kallikreins. Diabetic mice show hyperglycemia and increased diuresis, marked proteinuria and increased excretion of nitrites and kallikreins. B₂ receptor antagonists reduce water and protein excretion, compared to STZ group; STZ mice treated with B₁ receptor antagonists show normal glycemia and normalization of diuresis, protein, nitrite and kallikrein excretion.

The contractile response of the portal vein (a suitable preparation for B₁-BK studies) obtained from untreated 8-week old spontaneously hypertensive rats (SHR), is exaggerated and susceptible to enhanced capillary hydrostatic pressure and plasma leakage. Desendothelialized portal vein segments obtained from SHR are mounted in organ baths containing a Krebs solution for isometric contraction studies (baseline tension: 0.5 g). Test compounds are administered on portal vein segments obtained from normal rats and SHR, to establish dose-response curves.

Bradykinin B₁ receptor binding in human tissue is determined by the method of Levesque *et al.* [*Immunopharmacology* 29, 141-147 (1995); and *Immunopharmacology* 28, 1-7 (1994)]. Human embryonic fibroblast cells from the IMR-90 line (available from ATCC as CCL 186) are grown in minimal essential medium as described by Menke *et al.* [*J.Biol.Chem.* 269, 21583-21586 (1994)]. After 24 hours, the culture medium is replaced with low serum media (0.4% fetal bovine serum) containing recombinant human IL-1β (0.25 mg/mL) and the cells are

further incubated for 4-5 hours. The cells are harvested with trypsin and resuspended in Medium 11995-065 (Gibco, Gaithersburg, MD, USA) supplemented with L-glutamine, non-essential amino acids and 10% fetal bovine serum at 1.7×10^6 cells/mL. Thirty microliters of the cell suspension in a plate is mixed with 10 μ L of straight buffer [1 L of Medium 199 (Gibco, Gaithersburg, MD, USA), 25 mL of HEPES buffer, 1 g bovine serum albumin 3 μ M amastatin, 1 μ M captopril and 1 μ M phosphoramidon (Sigma, St. Louis, MO, USA)] or 10 μ L of buffer containing 5 to 50 μ M B₁-BK antagonist and 10 μ L of 11 μ M ³H-desArg¹⁰-kallidin. The plates are incubated at room temperature for about 1.5 hours. After incubation, each well is washed with 150 μ L of ice-cold PBS at pH 2.4. The contents are transferred to a glass fiber plate that has been pretreated with polyethyleneimine and the plate is air dried. Scintillation fluid is added and the resulting solution is counted in a gamma counter for 10 minutes. Statistical analysis is performed on the saturation curves. Scatchard regression parameters are calculated from the mean saturation data using a computer program (Tallarida and Murray, 1987). The resulting B_{max} and K_d values and their respective SEM are compared in order to assess statistical differences using Student's *t*-test. The compounds of the invention exhibit K_i's below 10 μ M. Specific examples of compounds exhibiting such activity are shown in Table 1.

Potency and efficacy in human tissue are assessed as follows: Human umbilical cords are obtained within 24 hours following normal deliveries and are stored in physiological salt solution (PSS) at 4° C. The composition of the PSS is as follows: 118mM NaCl, 4.6 mM KCl, 1.2 mM KH₂PO₄, 1.2 mM MgSO₄, 2.5mM CaCl₂, 0.026 mM CaNa₂EDTA, 10 mM glucose, and 24.8 mM NaHCO₃. The umbilical vein is carefully dissected and placed in ice-cold, PSS, which is continuously aerated with 95% O₂/5%CO₂ to maintain pH at 7.4. Excess connective tissue is removed, and rings 2-3 mm in length are prepared. The rings are mounted between stainless steel wires in water-jacketed tissue baths for

measuring contractile function. The rings are attached to a force-displacement transducer for measuring tension development. The baths contain 15 mL of oxygenated PSS maintained at 37° C.

- After mounting, resting tension is adjusted to 1.0 g and the rings are
- 5 equilibrated for 60 minutes before beginning the experiment. The tissue baths are rinsed with fresh PSS 30 min and 60 min after mounting the rings. Following each rinse, the resting tension is adjusted to 1.0 g. After the equilibration period, the rings are depolarized by adding increasing concentrations of KCl to the tissue bath until a maximum increase in tension is obtained. The bath is rinsed with fresh PSS,
 - 10 and the resting tension readjusted to 1.0 g. the response to KCl is repeated two additional times at 30-min intervals. The maximum increases in tension obtained following the second and third assessments of the response to KCl are averaged. The value is used to normalize the direct response to the test compound, and also the response to a reference bradykinin receptor agonist.
 - 15 Evaluating an Antagonist Effect: After assessing the responses to KCl, the test compound is added to the tissue bath. Thirty minutes later, the following concentrations of desArg¹⁰ Kallidin are added to the tissue bath: 0.01, 0.03, 0.1, 0.3, 1, 3, 10, 30, 100 nM. The response to each concentration of desArg¹⁰ Kallidin is normalized as a percentage of the maximum constrictor response to KCl.
 - 20 Evaluating a Direct Effect: After assessing the responses to KCl, the following concentrations of the test compound are added to the tissue bath: 1, 3, 10, 30, 100, 300, 1000, 3000 and 10000 nM. Alternatively, an equivalent volume of the vehicle used to solubilize the test compound is added to the tissue baths. Each new concentration is added to the bath after the response to the previous concentration
 - 25 has reached equilibrium. If no response is obtained, the next concentration of test compound is added to the bath 15 min after the previous concentration.

While it may be possible for the compounds of formula (I) to be administered as the raw chemical, it is preferable to present them as a pharmaceutical composition. According to a further aspect, the present invention provides a pharmaceutical composition comprising a compound of formula (I) or a pharmaceutically acceptable salt or solvate thereof, together with one or more pharmaceutically carriers thereof and optionally one or more other therapeutic ingredients, as discussed below. The carrier(s) must be "acceptable" in the sense of being compatible with the other ingredients of the formulation and not deleterious to the recipient thereof.

The formulations include those suitable for oral, parenteral (including subcutaneous, intradermal, intramuscular, intravenous and intraarticular), rectal and topical (including dermal, buccal, sublingual and intraocular) administration. The most suitable route may depend upon the condition and disorder of the recipient. The formulations may conveniently be presented in unit dosage form and may be prepared by any of the methods well known in the art of pharmacy. All methods include the step of bringing into association a compound of the invention or a pharmaceutically acceptable salt or solvate thereof ("active ingredient") with the carrier which constitutes one or more accessory ingredients. In general, the formulations are prepared by uniformly and intimately bringing into association the active ingredient with liquid carriers or finely divided solid carriers or both and then, if necessary, shaping the product into the desired formulation.

Pharmaceutical formulations, particularly topical formulations, may additionally comprise steroidal anti-inflammatory drugs, which may include but are not limited to alclometasone dipropionate, amcinonide, beclamethasone dipropionate, betamethasone benzoate, betamethasone dipropionate, betamethasone valerate, budesonide, clobetasol propionate, clobetasone butyrate, desonide, desoxymethasone, diflorasone diacetate, diflucortolone valerate, flumethasone

pivalate, flucolorolone acetonide, fluocinolone acetonide, fluocinonide, fluocortin butyl, fluocortolone preparations, fluprednidene acetate, flurandrenolone, halcinonide, hydrocortisone, hydrocortisone acetate, hydrocortisone butyrate, methylprednisolone acetate, mometasone furoate and triamcinolone acetonide.

- 5 Pharmaceutical formulations may also additionally comprise steroidal anti-inflammatory drugs for oral administration. These may include but are not limited to finasteride, betamethasone and hydrocortisone.

- 10 Alternatively or additionally, pharmaceutical formulations may additionally comprise nonsteroidal anti-inflammatory drugs (NSAIDs), which may include but are not limited to aminoarylcarboxylic acids (fenamic acid NSAIDs), arylacetic acids, arylbutyric acids such as fenbufen, arylpropionic acids (profens), pyrazoles such as epirizole, pyrazolones such as phenylbutazone, salicylic acids such as aspirin, oxicams and other compound classes that may be considered as NSAIDs including leucotriene antagonists. These formulations exhibit both the additive
- 15 effects of the individual components and synergistic effects from blocking of multiple pathways in the pain and inflammation pathway.

- 20 Propionic acid NSAIDs are non-narcotic analgesics/nonsteroidal antiinflammatory drugs having a free $-\text{CH}(\text{CH}_3)\text{COOH}$ group, which optionally can be in the form of a pharmaceutically acceptable salt group, e.g., $-\text{CH}(\text{CH}_3)\text{COO}^-$ Na^+ . The propionic acid side chain is typically attached directly or via a carbonyl function to a ring system, preferably to an aromatic ring system. Exemplary propionic acid NSAIDs include: ibuprofen, indoprofen, ketoprofen, naproxen, benoxaprofen, flurbiprofen, fenoprofen, pirofen, carprofen, oxaprozin, pranoprofen, miroprofen, tioxaprofen, suprofen, alminoprofen, tiaprofen, fluprofen,
- 25 and bucloxic acid. Structurally related propionic acid derivatives having similar

analgesic and antiinflammatory properties are also intended to be included in this group. Profens, as well as NSAIDs from other classes, may exhibit optical isomerism. The invention contemplates the use of pure enantiomers and mixtures of enantiomers, including racemic mixtures, although the use of the substantially
5 optically pure enantiomer will generally be preferred.

Acetic acid NSAIDs are non-narcotic analgesics/nonsteroidal antiinflammatory drugs having a free $-\text{CH}_2\text{COOH}$ group (which optionally can be in the form of a pharmaceutically acceptable salt group, e.g. $-\text{CH}_2\text{COO}^-\text{Na}^+$, typically attached directly to a ring system, preferably to an aromatic or
10 heteroaromatic ring system. Exemplary acetic acid NSAIDs include: ketorolac, indomethacin, sulindac, tolmetin, zomepirac, diclofenac, fenclofenac, alclofenac, ibufenac, isoxepac, furofenac, tiopinac, zidometacin, acemetacin, fentiazac, clidanac, oxpinac, and fenclozic acid. Structurally related acetic acid derivatives having similar analgesic and antiinflammatory properties are also intended to be
15 encompassed by this group.

Fenamic acid NSAIDs are non-narcotic analgesics/nonsteroidal antiinflammatory drugs having a substituted N-phenylanthranilic acid structure. Exemplary fenamic acid derivatives include mefenamic acid, meclofenamic acid, flufenamic acid, niflumic acid, and tolfenamic acid.

20 Biphenylcarboxylic acid NSAIDs are non-narcotic analgesics/nonsteroidal antiinflammatory drugs incorporating the basic structure of a biphenylcarboxylic acid. Exemplary biphenyl-carboxylic acid NSAIDs include diflunisal and flufenisal.

Oxicam NSAIDs are N-aryl derivatives of 4-hydroxyl-1,2-benzothiazine

1,1-dioxide-3-carboxamide. Exemplary oxicam NSAIDs are piroxicam, tenoxicam
sudoxicam and isoxicam.

Pharmaceutical formulations may also include cyclo-oxygenase (COX)
inhibitors (including arylpropionic acids such as ibuprofen and salicylic acids such
as aspirin), selective cyclooxygenase-1 (COX-1) inhibitors or selective cyclo-
oxygenase-2 (COX-2) inhibitors such as rofecoxib or celecoxib. These
formulations also exhibit both the additive effects of the individual components
and synergistic effects from blocking of multiple pathways in the pain and
inflammation pathway.

The term "pharmaceutically acceptable salt" refers to salts prepared from
pharmaceutically acceptable non-toxic acids or bases including inorganic acids and
bases and organic acids and bases. When the compounds of the present invention
are basic, salts may be prepared from pharmaceutically acceptable non-toxic acids
including inorganic and organic acids. Suitable pharmaceutically acceptable acid
addition salts for the compounds of the present invention include acetic,
benzenesulfonic (besylate), benzoic, camphorsulfonic, citric, ethenesulfonic,
fumaric, gluconic, glutamic, hydrobromic, hydrochloric, isethionic, lactic, maleic,
malic, mandelic, methanesulfonic, mucic, nitric, pamoic, pantothenic, phosphoric,
succinic, sulfuric, tartaric acid, p-toluenesulfonic, and the like. When the
compounds contain an acidic side chain, suitable pharmaceutically acceptable base
addition salts for the compounds of the present invention include metallic salts
made from aluminum, calcium, lithium, magnesium, potassium, sodium and zinc
or organic salts made from lysine, N,N'-dibenzylethylenediamine, chlorprocaine,
choline, diethanolamine, ethylenediamine, meglumine (N-methylglucamine) and
procaine.

Formulations of the present invention suitable for oral administration may be presented as discrete units such as capsules, cachets or tablets each containing a predetermined amount of the active ingredient; as a powder or granules; as a solution or a suspension in an aqueous liquid or a non-aqueous liquid; or as an oil-in-water liquid emulsion or a water-in-oil liquid emulsion. The active ingredient may also be presented as a bolus, electuary or paste. A tablet may be made by compression or moulding, optionally with one or more accessory ingredients. Compressed tablets may be prepared by compressing in a suitable machine the active ingredient in a free-flowing form such as a powder or granules, optionally mixed with a binder, lubricant, inert diluent, lubricating, surface active or dispersing agent. Molded tablets may be made by molding in a suitable machine a mixture of the powdered compound moistened with an inert liquid diluent. The tablets may optionally be coated or scored and may be formulated so as to provide sustained, delayed or controlled release of the active ingredient therein.

Formulations for parenteral administration include aqueous and non-aqueous sterile injection solutions which may contain anti-oxidants, buffers, bacteriostats and solutes which render the formulation isotonic with the blood of the intended recipient. Formulations for parenteral administration also include aqueous and non-aqueous sterile suspensions, which may include suspending agents and thickening agents. The formulations may be presented in unit-dose or multi-dose containers, for example sealed ampules and vials, and may be stored in a freeze-dried (lyophilized) condition requiring only the addition of a sterile liquid carrier, for example saline, phosphate-buffered saline (PBS) or the like, immediately prior to use. Extemporaneous injection solutions and suspensions may be prepared from sterile powders, granules and tablets of the kind previously described.

Formulations for rectal administration may be presented as a suppository with the usual carriers such as cocoa butter or polyethylene glycol. Formulations for topical administration in the mouth, for example buccally or sublingually, include lozenges comprising the active ingredient in a flavoured basis such as sucrose and acacia or tragacanth, and pastilles comprising the active ingredient in a basis such as gelatin and glycerin or sucrose and acacia. It should be understood that in addition to the ingredients particularly mentioned above, the formulations of this invention may include other agents conventional in the art having regard to the type of formulation in question, for example those suitable for oral administration may include flavoring agents.

Preferred unit dosage formulations are those containing an effective dose, as hereinbelow recited, or an appropriate fraction thereof, of the active ingredient. The compounds of the invention may be administered orally or via injection at a dose from 0.001 to 2500 mg/kg per day. The dose range for adult humans is generally from 0.005 mg to 10 g/day. Tablets or other forms of presentation provided in discrete units may conveniently contain an amount of compound of the invention which is effective at such dosage or as a multiple of the same, for instance, units containing 5 mg to 500 mg, usually around 10mg to 200mg.

The compounds of formula (I) are preferably administered orally or by injection (intravenous or subcutaneous). The precise amount of compound administered to a patient will be the responsibility of the attendant physician. However, the dose employed will depend on a number of factors, including the age and sex of the patient, the precise disorder being treated, and its severity. Also, the route of administration may vary depending on the condition and its severity.

EXAMPLE 1

Aqueous Suspension for Injection

A suspending vehicle is prepared from the following materials:

5	Polyethylene glycol 4000	30	gm.
	Potassium chloride	11.2	gm.
	Polysorbate 80	2	gm.
	Methylparaben	0.2	gm.
10	Water for injection q.s.	1000	mL.

The parabens are added to a major portion of the water and are dissolved therein by stirring and heating to 65° C. The resulting solution is cooled to room temperature and the remainder of the ingredients are added and dissolved. The balance of the water to make up the required volume is then added and the solution
 15 sterilized by filtration. The sterile vehicle thus prepared is then mixed with 3 gm of B₁-BK inhibitor of the invention (e.g. compound 10), which has been previously reduced to a particle size less than about 10 microns and sterilized with ethylene oxide gas. This mixture may then be mixed, optionally, with 5 gm of an antiinflammatory (e.g. hydrocortisone), which has been previously reduced to a
 20 particle size less than about 10 microns and sterilized with ethylene oxide gas. The mixture is passed through a sterilized colloid mill and filled under aseptic conditions into sterile containers which are then sealed.

EXAMPLE 2

Water-washable cream

The following ingredients are formulated:

5	Ingredients	Per Cent w/w
	Hydrocortisone acetate	0.025
	Compound 10	0.025
	Mineral Oil	6.0
10	Petrolatum	15.0
	Polyethylene glycol 1000 monocetyl ether	1.8
	Cetostearyl alcohol	7.2
	Chlorocresol	0.1
15	Distilled water to produce 100 parts by weight	

The cortisone and B_1 -BK antagonist **10** are ball-milled with a little mineral oil to a particle size of less than 5 microns. The water is heated to boiling, the chlorocresol added and the solution then cooled to 65° C. Then the petrolatum, cetostearyl alcohol and polyethylene glycol ether are mixed together while heating to 65° C. The milled steroid suspension is then added to the melt rinsing the container with mineral oil. The active ingredient oily phase thus prepared is added at 60° C to the chlorocresol aqueous phase at 65° C. The mixture is stirred rapidly while cooling past the gelling point (40° - 45° C.) and the stirring is continued at a speed sufficiently slow to permit the cream to set. The water-washable cream may be used in the treatment of dermatoses using either the open (without occlusion) or occlusive method of drug application.

EXAMPLE 3
Topical Ointment

5	Hydrocortisone acetate	0.05	gm
	Compound 10	1.00	gm.
	Chloroxine	1.00	gm.
	Propylene Glycol	7.00	gm.
	Glyceryl monostearate with emulsifier	5.00	gm.
10	White petrolatum q.s.a.d.	100.00	gm.

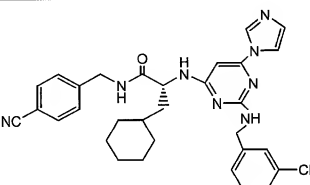
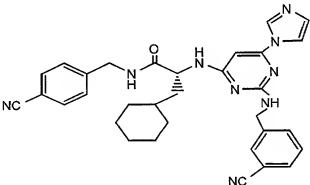
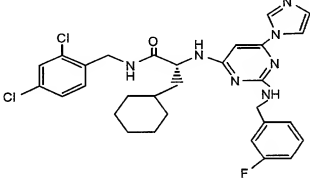
- Heat the propylene glycol to 55° C. Add hydrocortisone acetate, compound 10, and chloroxine and mix well. Add the remaining ingredients and mix until melted. Remove from heat and mix slowly until cooled to 45° C, then
- 15 homogenize.

EXAMPLE 4 - Tablets

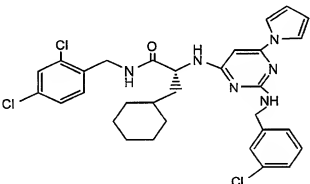
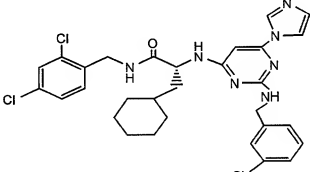
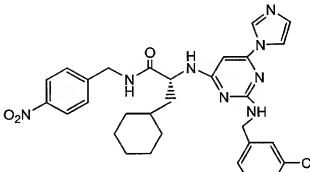
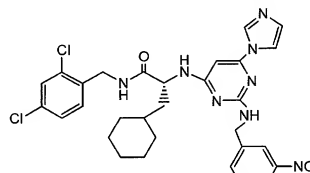
Composition per tablet:	
compound 10	30 mg
Precipitated calcium carbonate	50 mg
20 Corn Starch	40 mg
Lactose	73.4 mg
Hydroxypropylcellulose	6 mg
Magnesium stearate	(0.05 mL)
Total	200.0 mg

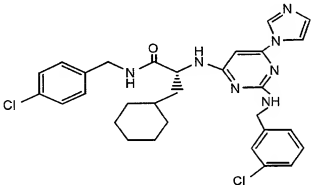
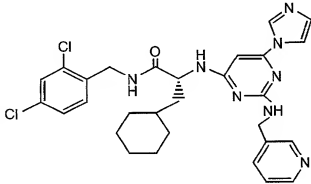
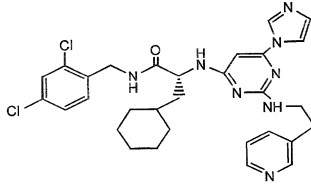
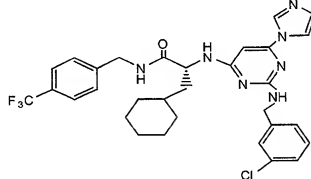
- 25 Compound **10**, precipitated calcium carbonate, corn starch, lactose and hydroxypropylcellulose are mixed together, water is added, and the mixture is kneaded, then dried in vacuum at 40°C for 16 hours, ground in a mortar and passed through a 16-mesh sieve to give granules. To this is added magnesium stearate and the resultant mixture is made up into tablets each weighing 200 mg on a rotary
- 30 tableting machine.

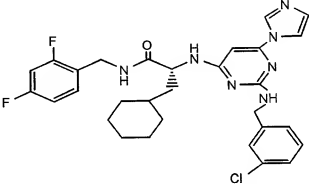
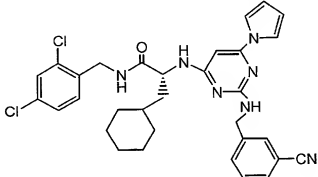
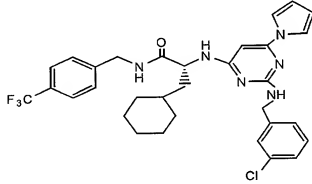
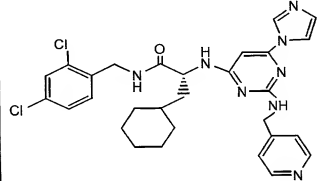
Table 1

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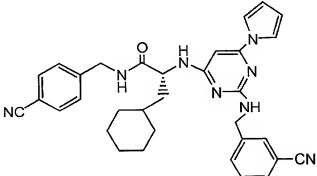
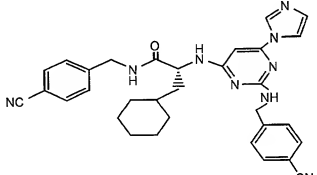
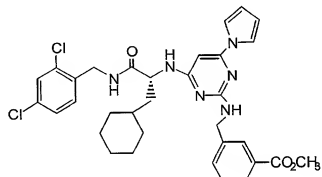
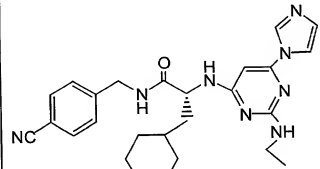
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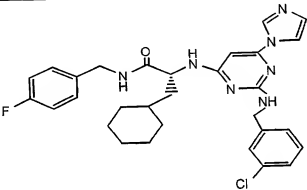
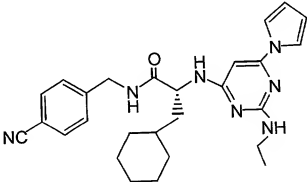
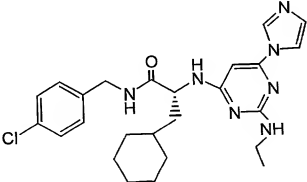
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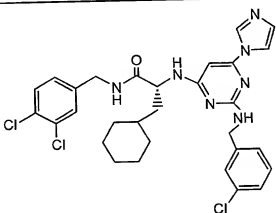
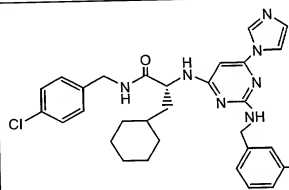
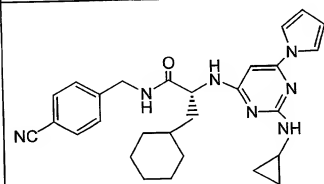
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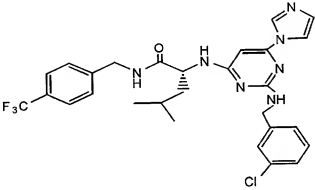
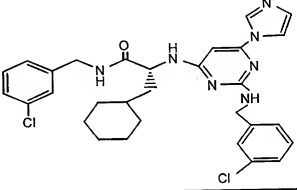
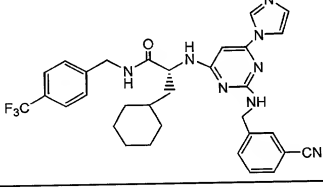
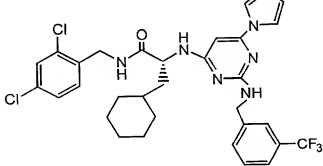
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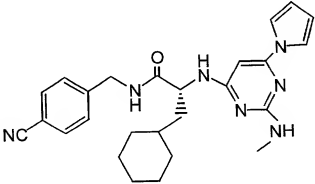
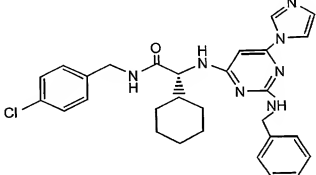
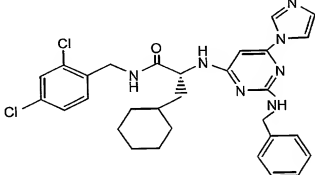
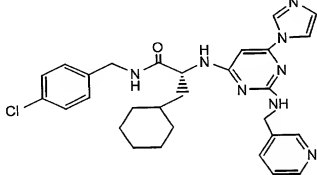
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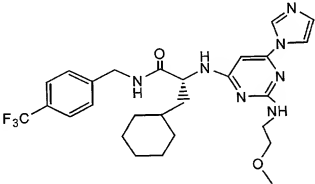
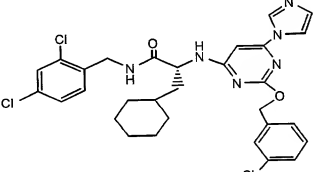
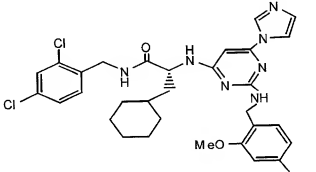
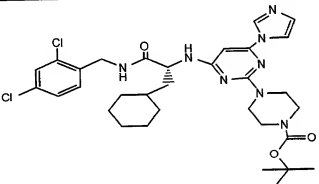
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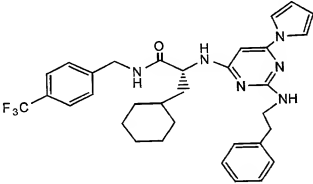
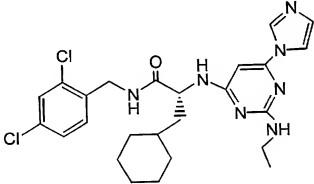
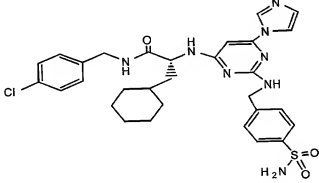
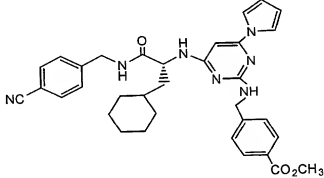
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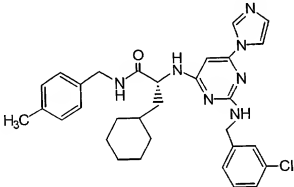
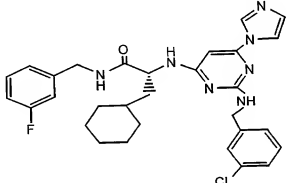
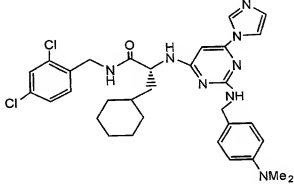
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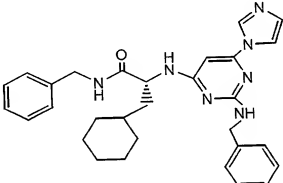
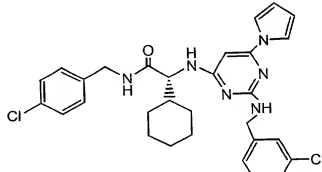
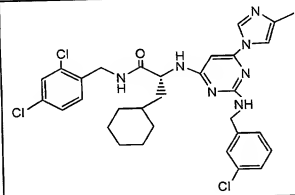
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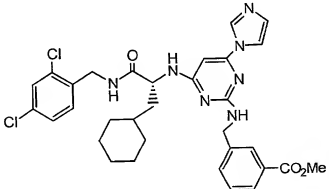
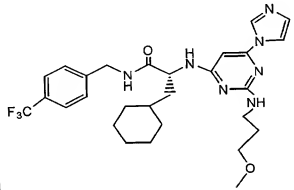
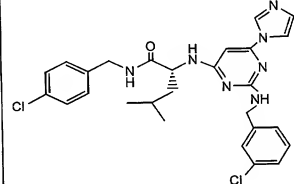
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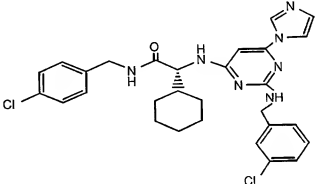
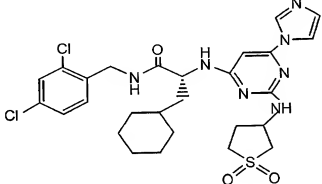
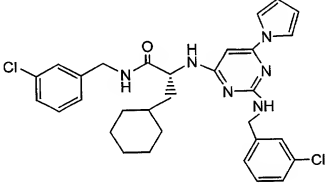
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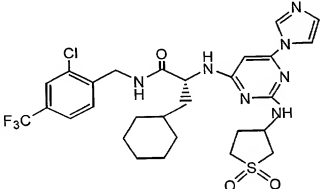
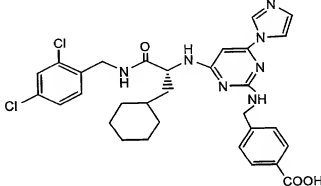
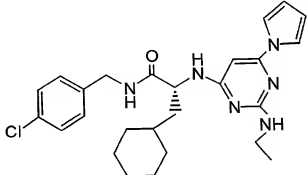
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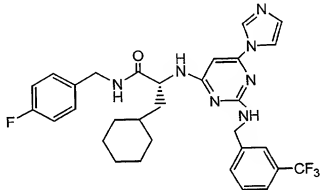
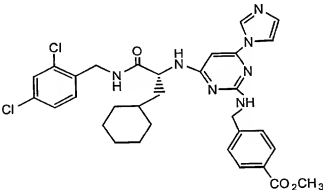
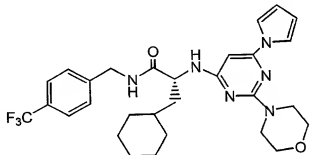
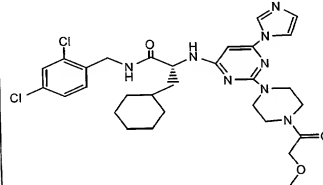
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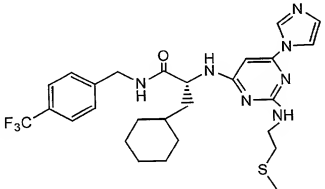
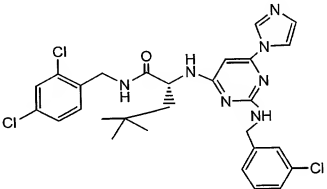
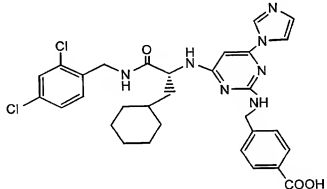
STRUCTURE	[M+H] ⁺
	510.00
	563
	626.00

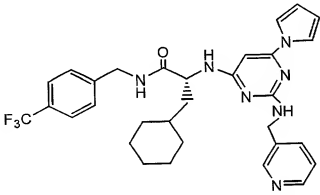
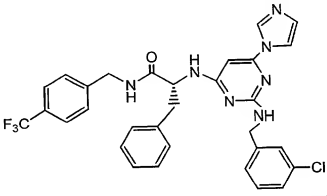
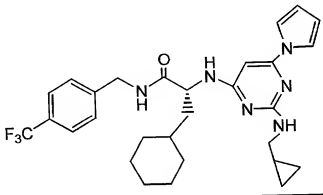
STRUCTURE	[M+H] ⁺
	636.6
	560
	538.00

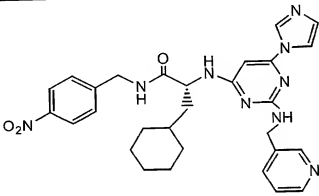
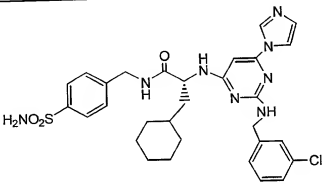
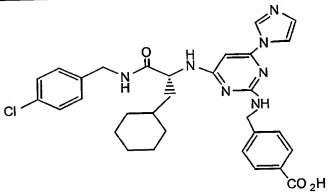
STRUCTURE	[M+H] ⁺
	564.00
	606.00
	563

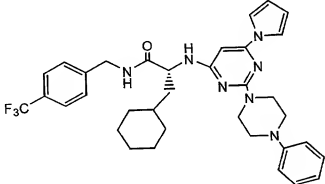
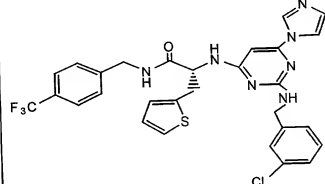
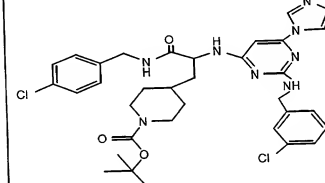
STRUCTURE	[M+H] ⁺
	606
	622
	481

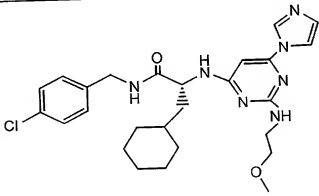
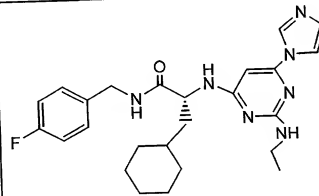
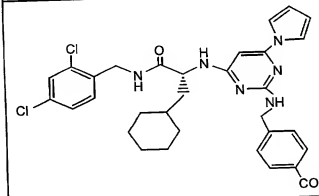
STRUCTURE	[M+H] ⁺
 <chem>CC1=NC2=C(N1)N=CN(C2)C(=N)N[C@@H](C3CCCCC3)C(=O)NCCc4ccc(F)cc4</chem>	596
 <chem>COC(=O)c1ccc(NC2=NC3=C(N2)N=CN(C3)C(=N)N[C@@H](C4CCCCC4)C(=O)NCCc5cc(Cl)cc(Cl)c5)cc1</chem>	636
 <chem>C1CCN(C1)c2nc3c(ncn3C(=N)N[C@@H](C4CCCCC4)C(=O)NCCc5ccc(C(F)(F)F)cc5)c4ccccc42</chem>	557.00
 <chem>COC(=O)N1CCN(C1)c2nc3c(ncn3C(=N)N[C@@H](C4CCCCC4)C(=O)NCCc5cc(Cl)cc(Cl)c5)c4ccccc42</chem>	629.00

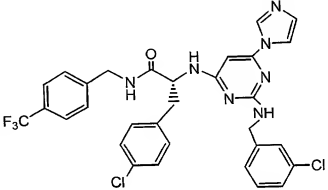
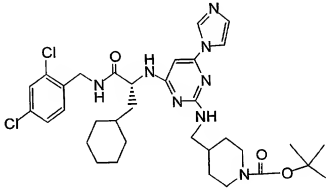
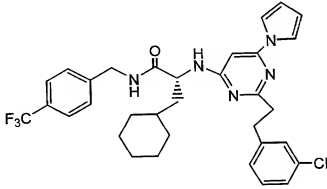
STRUCTURE	[M+H] ⁺
 <chem>CSc1nc2c(ncn2C3=CC=CC=C3C(F)(F)F)nc4c1cnc4C5=CC=CC=C5C6CCCCC6</chem>	562
 <chem>CC(C)(C)C[C@H](NC(=O)NC1=CC=C(C=C1)Cl)C2=CC=C(C=C2)Cl</chem>	586
 <chem>OC(=O)C1=CC=C(C=C1)N2C=NC3=C(N2)N=CN3C4=CC=CC=C4C5CCCCC5</chem>	622.00

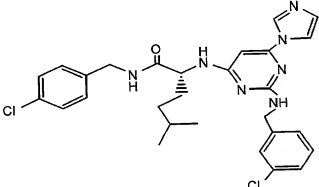
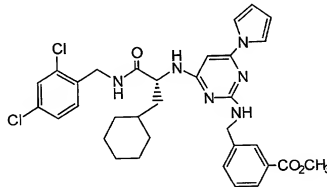
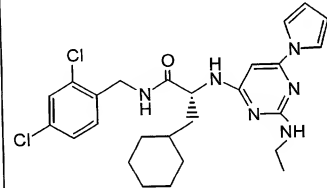
STRUCTURE	[M+H] ⁺
 <chem>CC1(C)CCCCC1[C@H](C(=O)Nc2cc(Cc3ccc(C(F)(F)F)cc3)nn2)Nc4nc(Cc5ccc(C(F)(F)F)cc5)nn4Cc6ccccn6</chem>	578.00
 <chem>CC1(C)CCCCC1[C@H](C(=O)Nc2cc(Cc3ccccc3)nn2)Nc4nc(Cc5cc(Cl)ccc5)nn4Cc6ccc(C(F)(F)F)cc6</chem>	606.00
 <chem>CC1(C)CCCCC1[C@H](C(=O)Nc2cc(Cc3C1CC3)nn2)Nc4nc(Cc5ccccc5)nn4Cc6ccc(C(F)(F)F)cc6</chem>	541.00

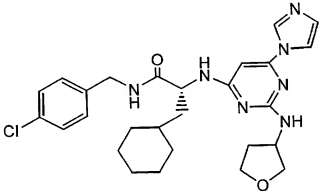
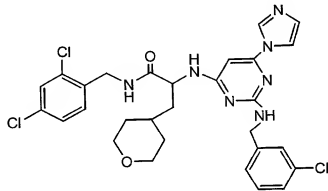
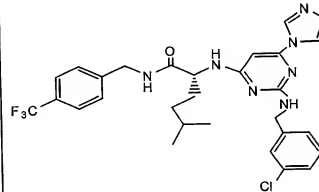
STRUCTURE	[M+H] ⁺
	556.00
	623
	588.00

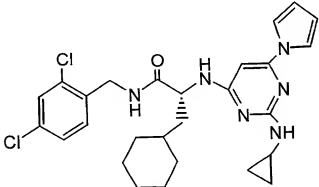
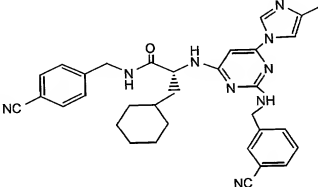
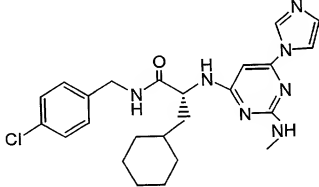
STRUCTURE	[M+H] ⁺
 <chem>CC1(C)CC(C1)C(=O)NCCc2ccc(C(F)(F)F)cc2[C@H](C3CCCCC3)NC(=N)C4=NC5=CC=CC=C5N4CN6CCCCC6</chem>	632.00
 <chem>Clc1ccc(cc1)CNC2=NC3=NC(=NC2N)N(C3)N4C=CC=C4S4C[C@H](C4)C(=O)NCCc5ccc(C(F)(F)F)cc5</chem>	612.00
 <chem>CC(C)(C)C(=O)OCC1CC[C@H](C1)C(=O)NCCc2ccc(Cl)cc2[C@H](C3CCCCC3)NC(=N)C4=NC5=CC=C(C=C5)N4CN6CCCCC6</chem>	679.00

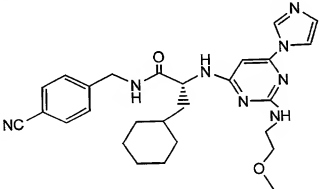
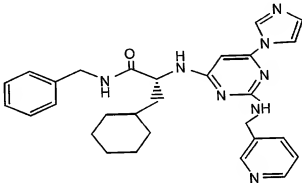
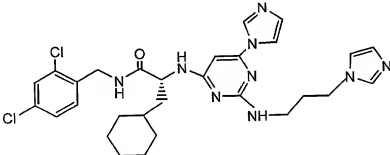
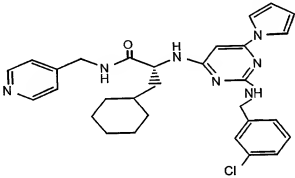
STRUCTURE	[M+H] ⁺
 <p>Chemical structure of a triazole derivative. It features a 1,2,4-triazole ring substituted with a 2-methoxyethyl group at position 4, a 1-cyclohexyl group at position 5, and a 1-((4-chlorobenzyl)amino)propan-2-yl group at position 3.</p>	512
 <p>Chemical structure of a triazole derivative. It features a 1,2,4-triazole ring substituted with an ethyl group at position 4, a 1-cyclohexyl group at position 5, and a 1-((4-fluorobenzyl)amino)propan-2-yl group at position 3.</p>	466
 <p>Chemical structure of a triazole derivative. It features a 1,2,4-triazole ring substituted with a 4-carboxybenzyl group at position 4, a 1-cyclohexyl group at position 5, and a 1-((3,4-dichlorobenzyl)amino)propan-2-yl group at position 3.</p>	621

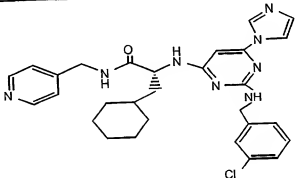
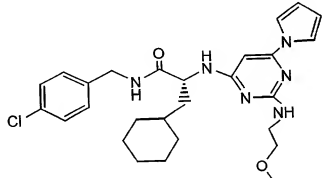
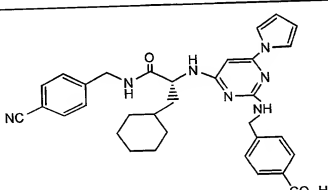
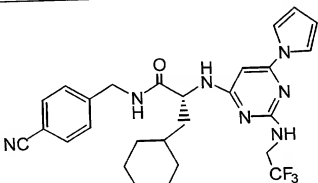
STRUCTURE	[M+H] ⁺
	634.00
	685
	610.00

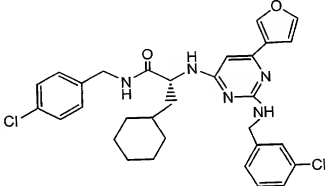
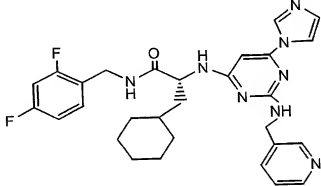
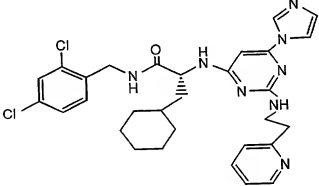
STRUCTURE	[M+H] ⁺
	586.56
	635
	515

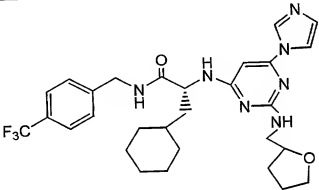
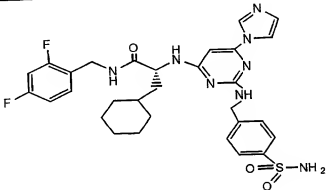
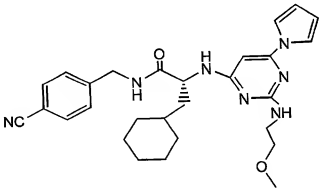
STRUCTURE	[M+H] ⁺
	524
	614
	586.00

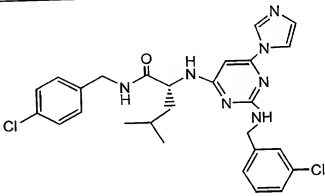
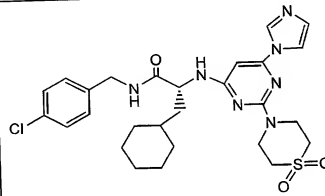
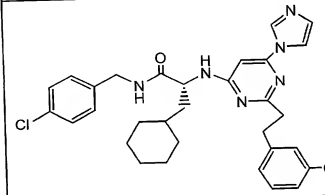
STRUCTURE	[M+H] ⁺
	527
	574
	468.00

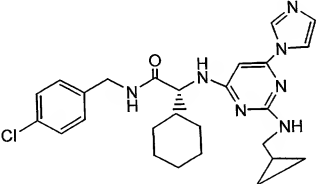
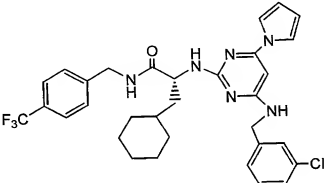
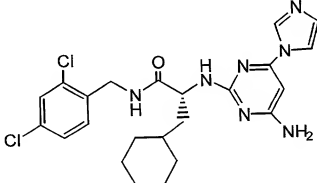
STRUCTURE	[M+H] ⁺
	503
	511.00
	596.00
	544.00

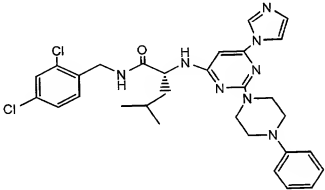
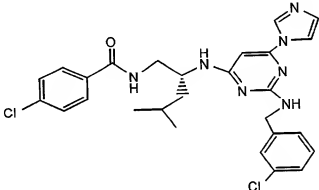
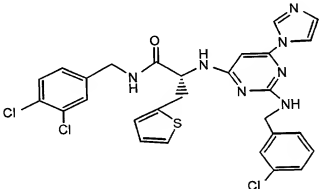
STRUCTURE	[M+H] ⁺
	545.00
	511
	578
	526

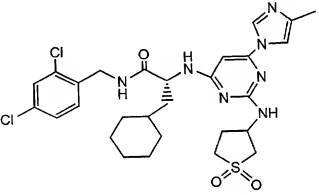
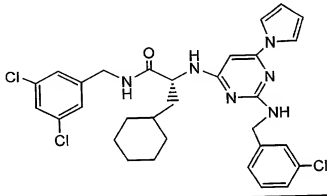
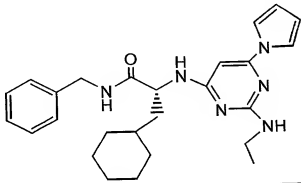
STRUCTURE	[M+H] ⁺
	578.00
	547.00
	593.00

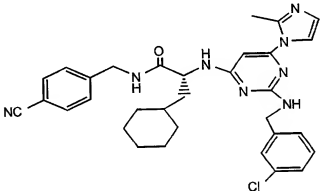
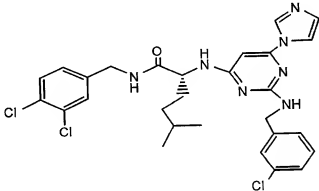
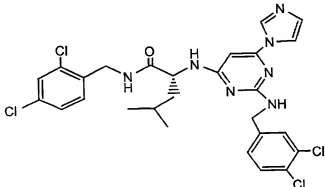
STRUCTURE	[M+H] ⁺
	572
	625.00
	502

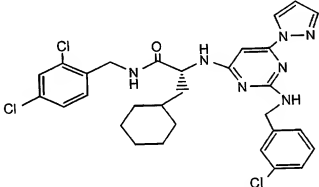
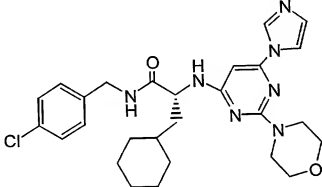
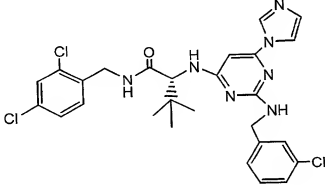
STRUCTURE	[M+H] ⁺
 <chem>CC(C)[C@H](NC(=O)NCc1ccc(Cl)cc1)Nc2nc3cc(Cc4ccc(Cl)cc4)nn3n2</chem>	538.53
 <chem>O=S1(=O)CN(C1)c2nc3cc(Cc4ccc(Cl)cc4)nn3n2</chem>	572
 <chem>Clc1ccc(CCCc2nc3cc(Cc4ccc(Cl)cc4)nn3n2)cc1</chem>	577.00

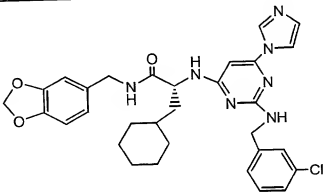
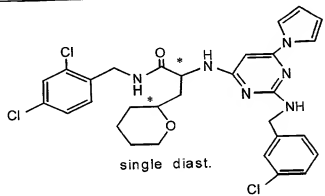
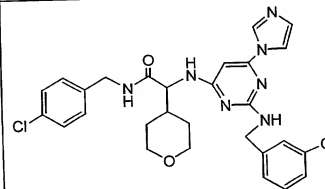
STRUCTURE	[M+H] ⁺
 <chem>Clc1ccc(cc1)CN(C(=O)N[C@@H](C2CCCCC2)Nc3nc(NC4CC4)c5ccn5n3)c6ccc(cc6)N7C=CN=C7</chem>	494
 <chem>Fc1ccc(cc1)CN(C(=O)N[C@@H](C2CCCCC2)Nc3nc(NC4C=CC=CC4Cl)c5ccn5n3)c6ccc(cc6)N7C=CC=CC7</chem>	611.00
 <chem>Clc1cc(Cl)ccc1CN(C(=O)N[C@@H](C2CCCCC2)Nc3nc(NC4C=CC=C4N)c5ccn5n3)c6ccc(cc6)N7C=CC=CC7</chem>	488.00

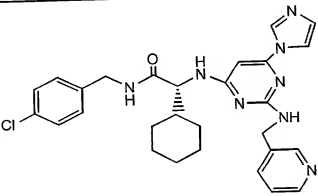
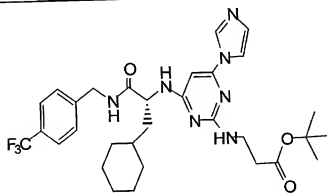
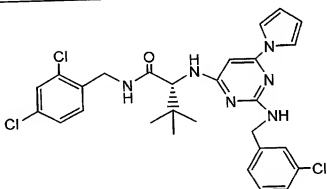
STRUCTURE	[M+H] ⁺
	593.00
	538.00
	612.00

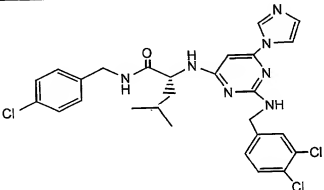
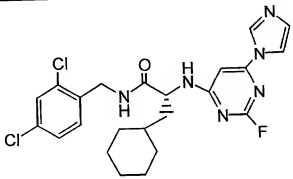
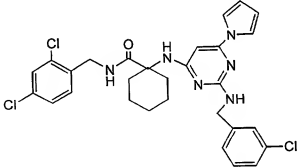
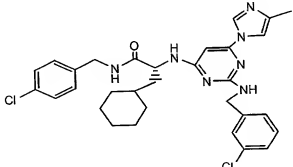
STRUCTURE	[M+H] ⁺
 <chem>CC1=CN=C2C(=N1)N(C(=N2)N[C@@H](C3CCCCC3)C(=O)NCc4cc(Cl)cc(Cl)c4)N[C@@H]5CCSC5=O</chem>	620
 <chem>Clc1ccc(CN[C@@H](C2CCCCC2)C(=O)NCc3cc(Cl)cc(Cl)c3)c4nn5c(NC6=CC=CC=C6C)ncn5c4</chem>	625
 <chem>CC1=CN=C2C(=N1)N(C(=N2)N[C@@H](C3CCCCC3)C(=O)NCc4ccccc4)N[C@@H]5CCN5</chem>	447

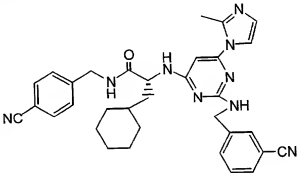
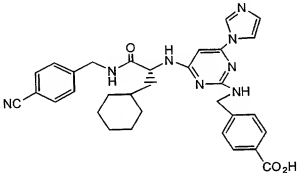
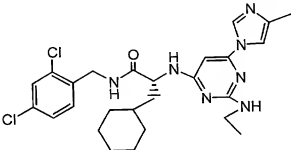
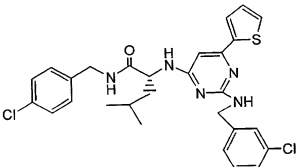
STRUCTURE	[M+H] ⁺
	583
	586.00
	606.00

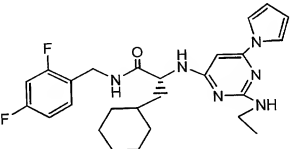
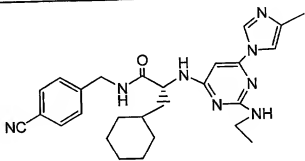
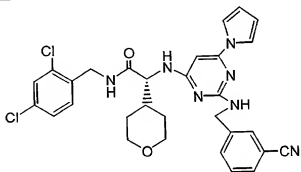
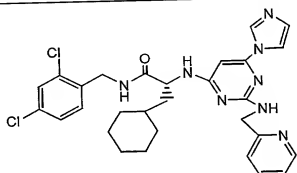
STRUCTURE	[M+H] ⁺
 <chem>Clc1ccc(cc1CN(C(=O)[C@H](C1CCCCC1)Nc2nc3ccnnc3cc2Nc4ccccc4Cl)C(=O)Nc5cc(Cl)cc(Cl)c5)N6C=CC=CC=N6</chem>	612.00
 <chem>Clc1ccc(cc1CN(C(=O)[C@H](C1CCCCC1)Nc2nc3ccnnc3cc2Nc4ccccc4N5CCOCC5)C(=O)Nc6cc(Cl)ccc6)N7C=CC=CC=N7</chem>	524.00
 <chem>Clc1ccc(cc1CN(C(=O)[C@H](C(C)(C)C)Nc2nc3ccnnc3cc2Nc4cc(Cl)ccc4)C(=O)Nc5cc(Cl)ccc5)N6C=CC=CC=N6</chem>	572

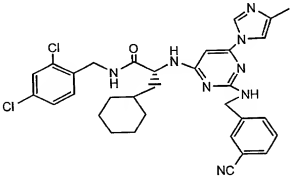
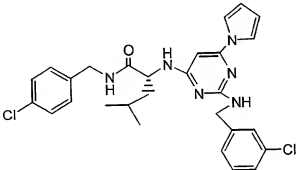
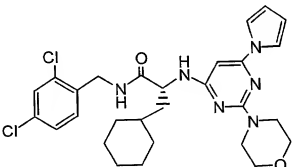
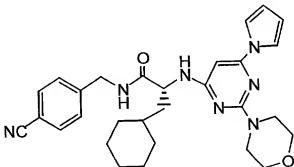
STRUCTURE	$[M+H]^+$
	588.00
 <p>single diast.</p>	613.00
	566.5

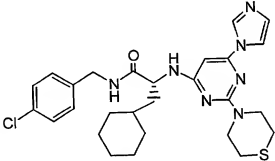
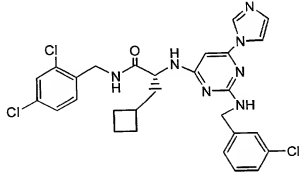
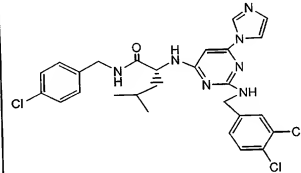
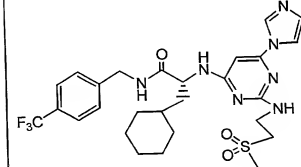
STRUCTURE	[M+H] ⁺
	531
	616
	571

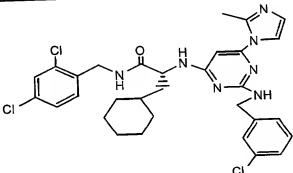
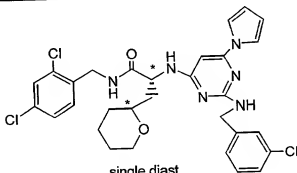
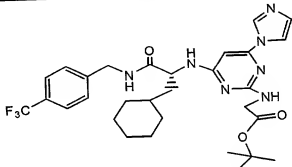
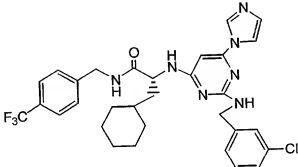
STRUCTURE	[M+H] ⁺
	572.00
	491.00
	583.00
	592.00

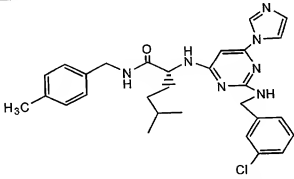
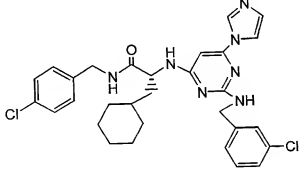
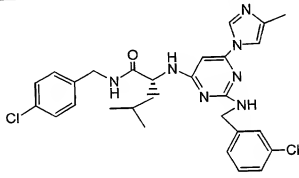
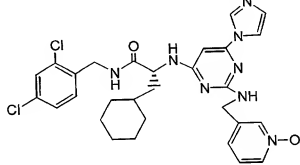
STRUCTURE	[M+H] ⁺
	574
	579
	530
	554.00

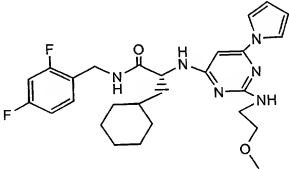
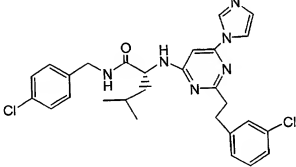
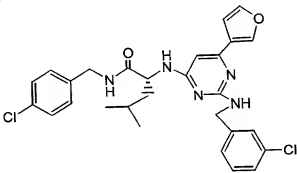
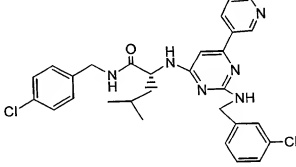
STRUCTURE	[M+H] ⁺
	483
	487
	590
	579.00

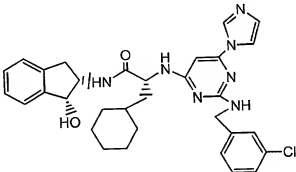
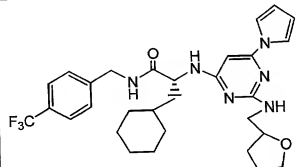
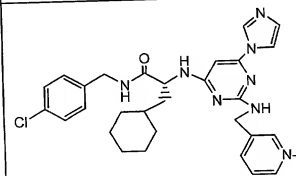
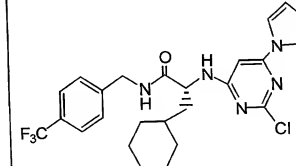
STRUCTURE	[M+H] ⁺
	617
	537.00
	557
	514

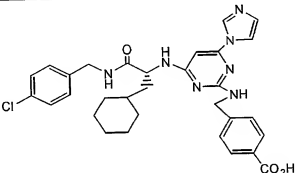
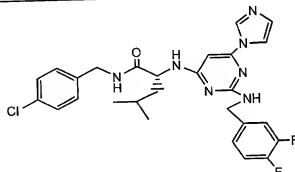
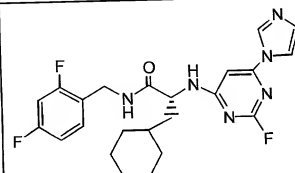
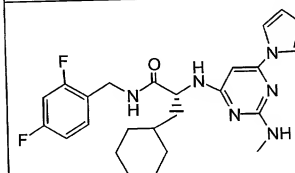
STRUCTURE	[M+H] ⁺
	540
	584
	572.00
	594

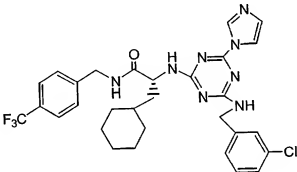
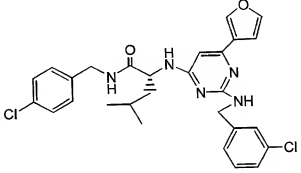
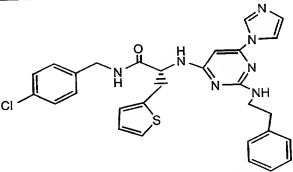
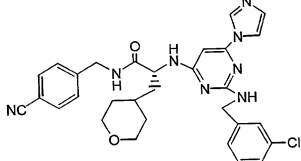
STRUCTURE	[M+H] ⁺
	626.00
 <p>single diast.</p>	614.00
	602
	612

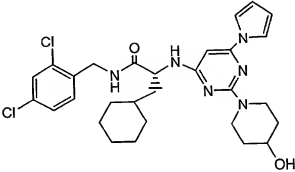
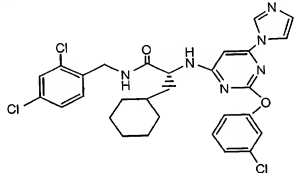
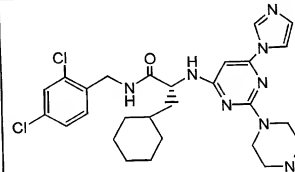
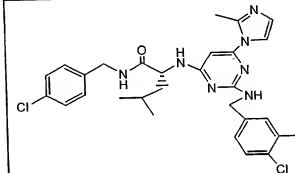
STRUCTURE	[M+H] ⁺
	532.00
	578
	552.00
	595

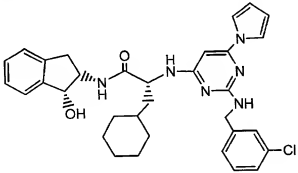
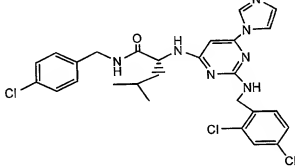
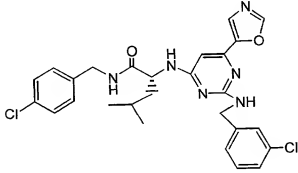
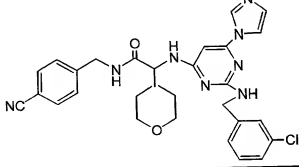
STRUCTURE	[M+H] ⁺
	513
	537.00
	538.00
	549.00

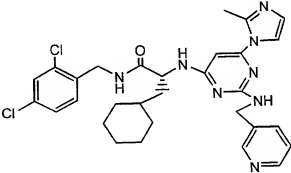
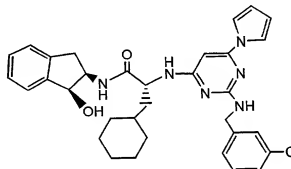
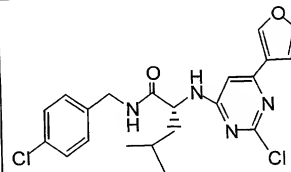
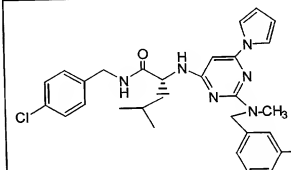
STRUCTURE	[M+H] ⁺
	586
	571
	561
	506.00

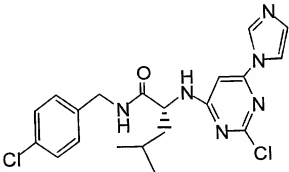
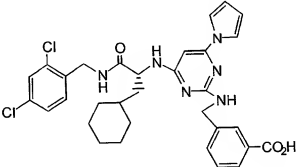
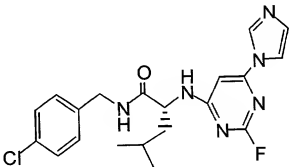
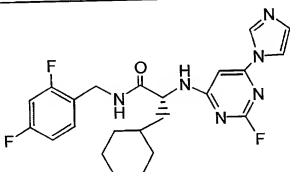
STRUCTURE	[M+H] ⁺
	588
	540.00
	459.00
	469

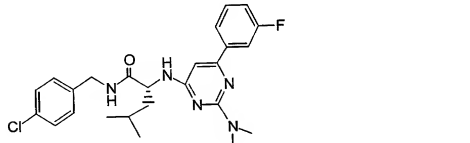
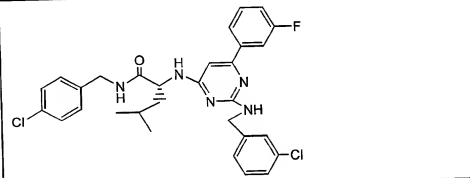
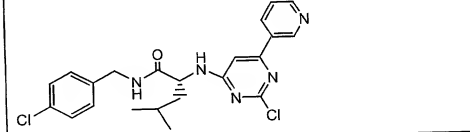
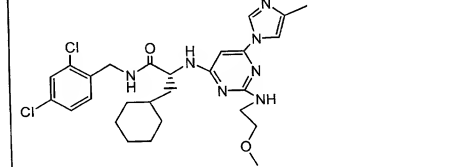
STRUCTURE	[M+H] ⁺
	613
	538.00
	558.00
	571.1

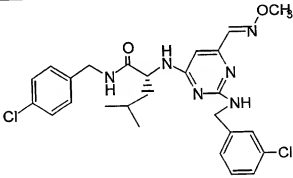
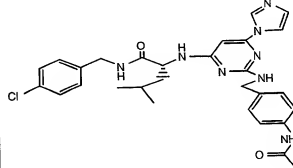
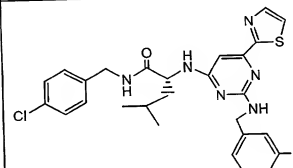
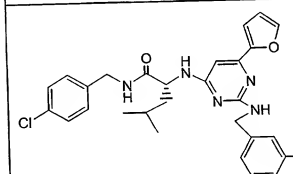
STRUCTURE	[M+H] ⁺
 <chem>Clc1cc(Cl)ccc1CN[C@@H](C1CCCCC1)C(=O)Nc2nc3c(ncn3C4CCCCC4O)nc5ccccc5n2</chem>	571
 <chem>Clc1cc(Cl)ccc1CN[C@@H](C1CCCCC1)C(=O)Nc2nc3c(ncn3Oc4ccccc4Cl)nc5ccccc5n2</chem>	599.00
 <chem>Clc1cc(Cl)ccc1CN[C@@H](C1CCCCC1)C(=O)Nc2nc3c(ncn3C4CCCCC4N)nc5ccccc5n2</chem>	557.00
 <chem>Clc1ccc(cc1)CN[C@@H](C(C)C)C(=O)Nc2nc3c(ncn3Cc4cc(Cl)cc(Cl)c4)nc5ccccc5n2</chem>	586.00

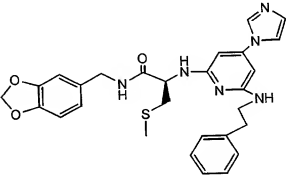
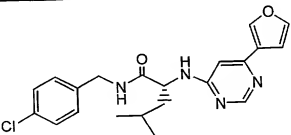
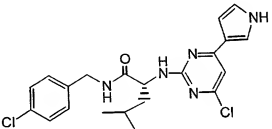
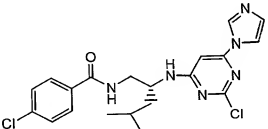
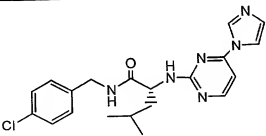
STRUCTURE	[M+H] ⁺
	585
	572.00
	539.00
	557

STRUCTURE	[M+H] ⁺
	593.00
	584
	433.00
	551.00

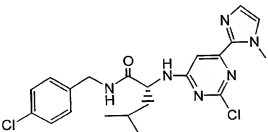
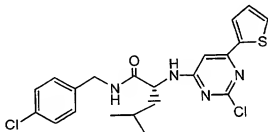
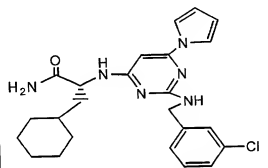
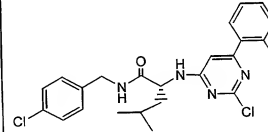
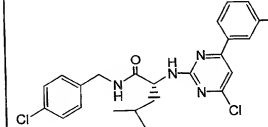
STRUCTURE	[M+H] ⁺
 <chem>CC(C)[C@H](NC(=O)NC1=CC=C(C=C1)Cl)Nc2cc(Cl)nnc2c3ccncc3</chem>	433.00
 <chem>OC(=O)c1ccc(NC2=CC=C(C=C2)Nc3cc(Cl)nnc3[C@H](C4CCCCC4)NC(=O)NC5=CC(=CC(=C5)Cl)Cl)cc1</chem>	621
 <chem>CC(C)[C@H](NC(=O)NC1=CC=C(C=C1)Cl)Nc2cc(F)nnc2c3ccncc3</chem>	417.00
 <chem>Fc1cc(F)ccc1CNc2cc(F)nnc2[C@H](C3CCCCC3)NC(=O)NC4=CC=C(C=C4)F</chem>	459

STRUCTURE	[M+H] ⁺
	470.00
	566.00
	444.00
	560.00

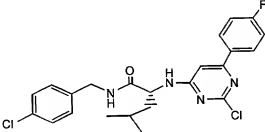
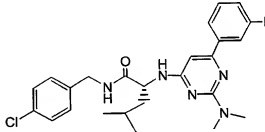
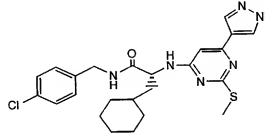
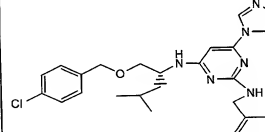
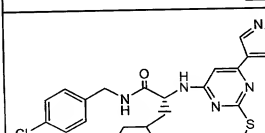
STRUCTURE	[M+H] ⁺
	529.00
	561.00
	555.00
	538.00

STRUCTURE	[M+H] ⁺
	532.00
	399.00
	432.00
	433.00
	439.00

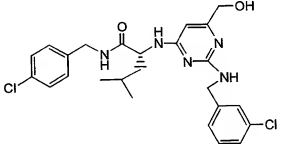
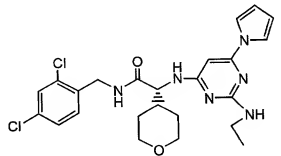
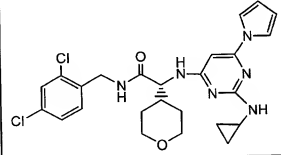
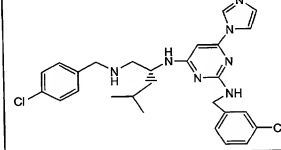
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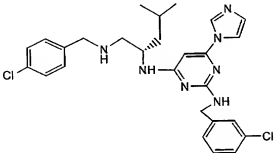
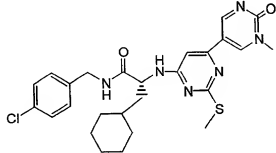
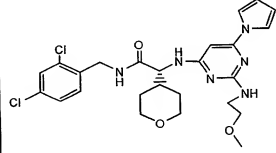
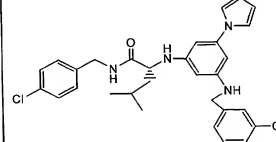
STRUCTURE	[M+H] ⁺
	447.00
	449.00
	453
	461.00
	461.00

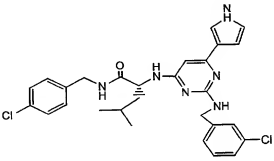
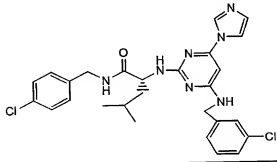
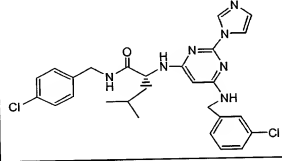
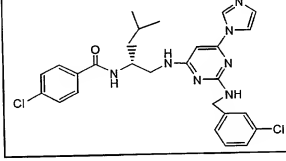
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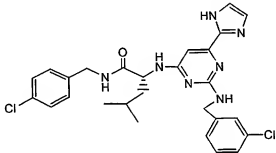
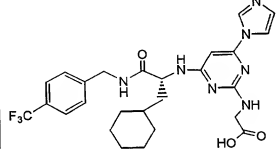
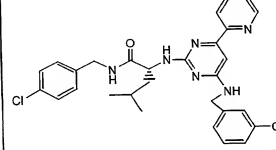
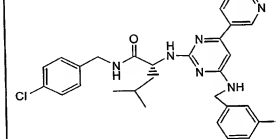
STRUCTURE	[M+H] ⁺
	461.00
	470.00
	484.00
	491.00
	499.00

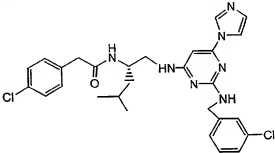
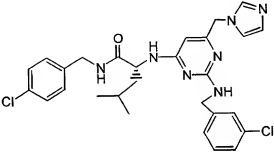
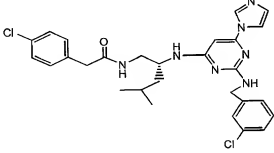
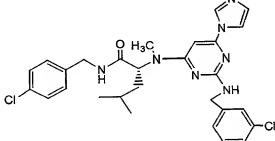
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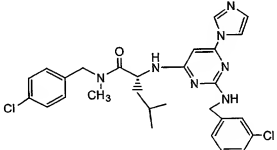
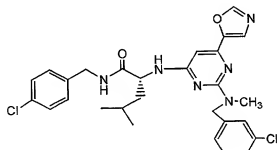
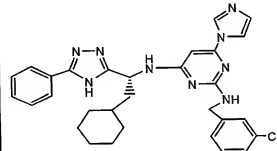
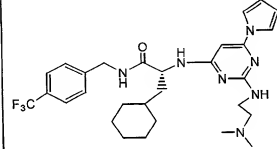
STRUCTURE	[M+H] ⁺
	502.00
	503
	515
	524.00

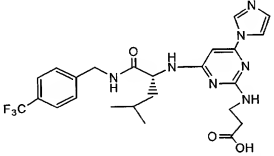
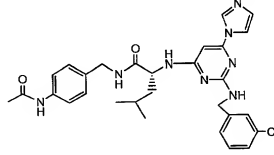
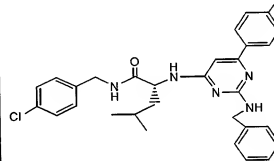
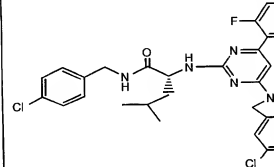
STRUCTURE	[M+H] ⁺
	524.00
	527.00
	533
	535

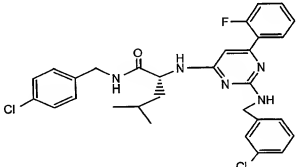
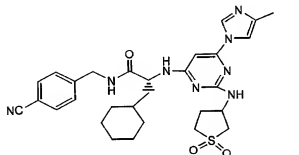
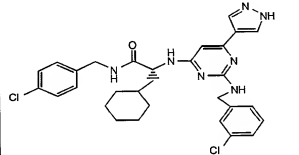
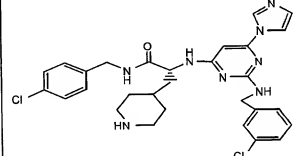
STRUCTURE	[M+H] ⁺
	537.00
	538.00
	538.00
	538.00

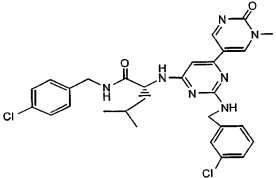
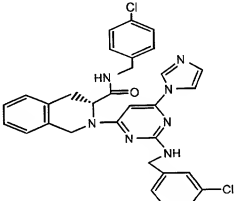
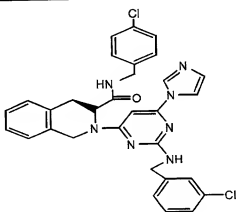
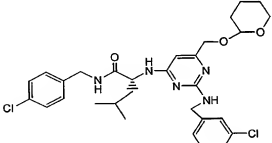
STRUCTURE	[M+H] ⁺
	538.00
	546
	549.00
	549.00

STRUCTURE	[M+H] ⁺
	552
	552.00
	552.00
	552.00

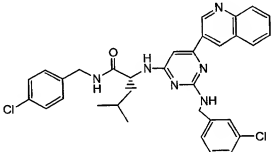
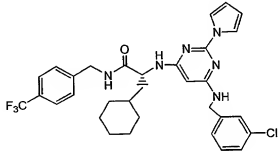
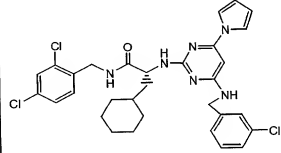
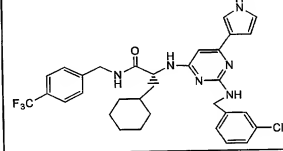
STRUCTURE	[M+H] ⁺
	552.00
	553.00
	554
	558.00

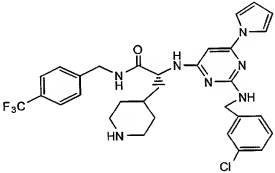
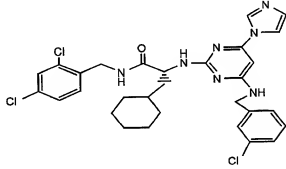
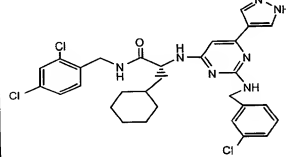
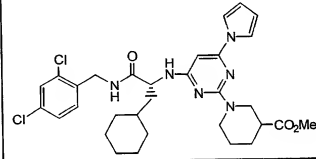
STRUCTURE	[M+H] ⁺
	560
	561.00
	566.00
	566.00

STRUCTURE	[M+H] ⁺
	566.00
	577
	578.00
	579.00

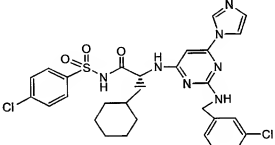
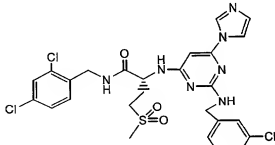
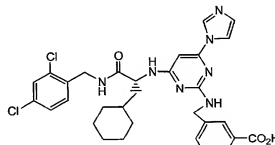
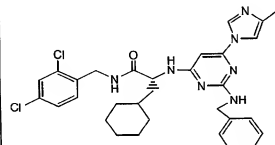
STRUCTURE	[M+H] ⁺
	580.00
	584.00
	584.00
	586.00

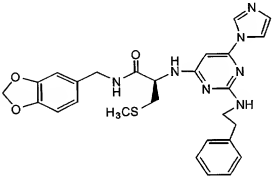
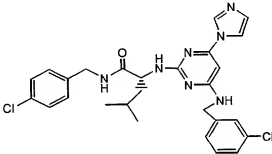
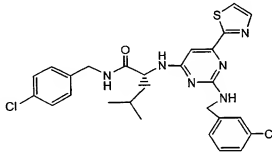
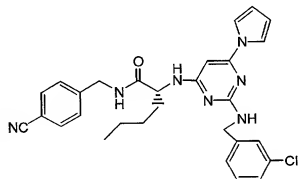
STRUCTURE	[M+H] ⁺
	588
	590.00
	592.00
	592.00

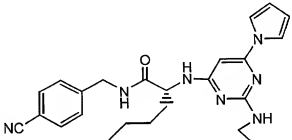
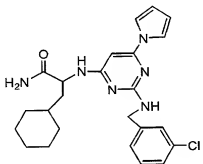
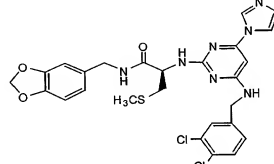
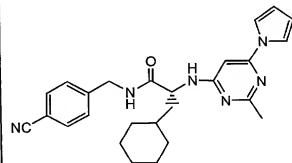
STRUCTURE	[M+H] ⁺
	599.00
	611.00
	611.00
	611.00

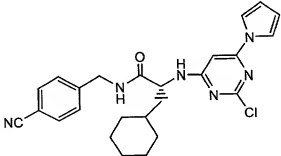
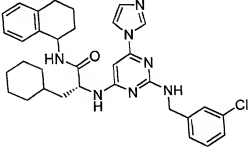
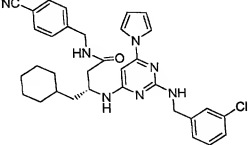
STRUCTURE	[M+H] ⁺
	612.00
	612.00
	612.00
	613

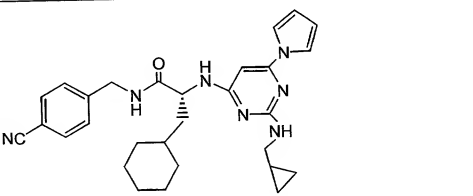
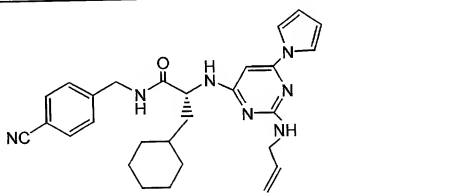
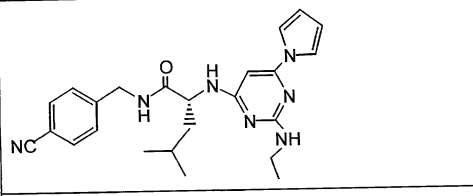
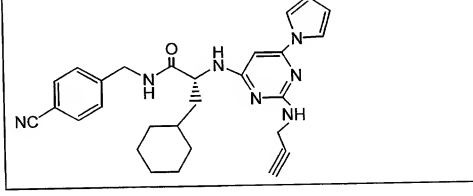
1073.035A.03402

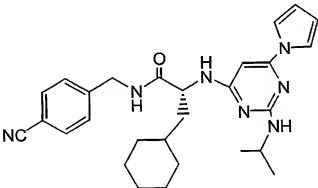
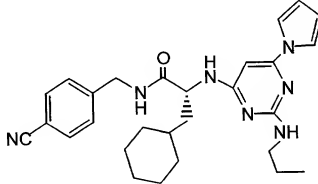
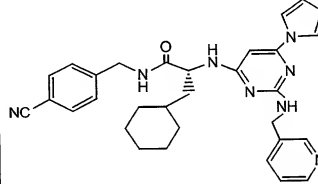
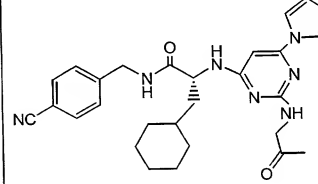
STRUCTURE	[M+H] ⁺
	614
	622
	622
	636.00

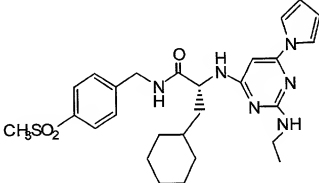
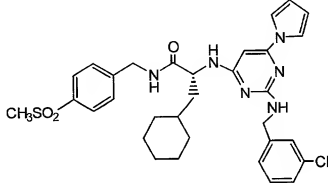
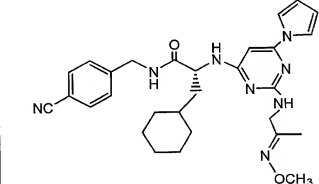
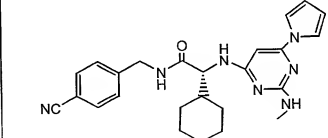
STRUCTURE	[M+H] ⁺
	532.00
	538.00
	555.00
	

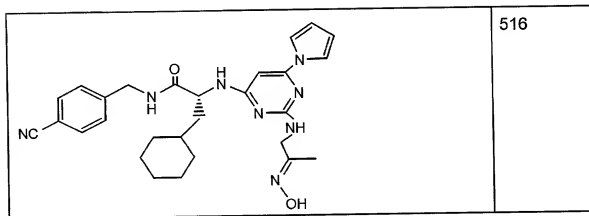
STRUCTURE	[M+H] ⁺
	
	
	586.00
	423

STRUCTURE	$[M+H]^+$
	463
	598
	582.3

	500
	486
	432
	482

 <chem>N#Cc1ccc(cc1)CN[C@@H](C2CCCCC2)C(=O)Nc3nc4c(ncn3C5C=CC=CC5)n4</chem>	486
 <chem>N#Cc1ccc(cc1)CN[C@@H](C2CCCCC2)C(=O)Nc3nc4c(ncn3C5C=CC=CC5)NCC4</chem>	486
 <chem>N#Cc1ccc(cc1)CN[C@@H](C2CCCCC2)C(=O)Nc3nc4c(ncn3C5C=CC=CC5)NCc6cccnc6</chem>	436
 <chem>CC(=O)C[C@@H](C2CCCCC2)C(=O)Nc3nc4c(ncn3C5C=CC=CC5)NCC4</chem>	501

	526
	621
	530
	444



1073.035A-014402